

# MENTAL APTITUDE

Questions : 50

**Direction for questions 1 - 10 :**

In the problem Figures, the first two figures bear a definite relationship with each other, and the third figure bears the same relation with one of the four figures given as Answer Figures, marked (A), (B), (C) and (D). Indicate the correct answer.

**Questions:**

**PROBLEM FIGURES**

**ANSWER FIGURES**

1.			A		B		C		D	
2.			A		B		C		D	
3.			A		B		C		D	
4.			A		B		C		D	
5.			A		B		C		D	
6.			A		B		C		D	
7.			A		B		C		D	
8.			A		B		C		D	
9.			A		B		C		D	
10.			A		B		C		D	



11. If + means  $\div$ ,  $\times$  means  $-$ ,  $\div$  means  $\times$  and  $-$  means  $+$ , then  $8 + 6 \times 4 \div 3 - 4 = ?$

- (A) - 12 (B) - 20/3  
(C) 12 (D) 20/3

**Direction for questions 12 - 14 :**

In each of the following questions, four groups of letters are given; three of them are alike in a certain way while one is different. Choose the ODD one.

**Questions :**

12. (A) FIL (B) RUX  
(C) ILO (D) LOQ  
13. (A) AZBY (B) PTQS  
(C) CWDW (D) GQHR  
14. (A) NEXFL (B) LANCP  
(C) FRGSP (D) ZGPKU

15. In a row of 60, A is standing at 10<sup>th</sup> from the right end, how many places should A move left ward to become 23<sup>rd</sup> from the left end ?

- (A) 25 (B) 26  
(C) 27 (D) 28

**Direction for questions 16 - 19 :**

Each question is followed by two statements (A) and (B)

**Indicate**

- (1) If statement (A) ALONE is sufficient, but statement (B) alone is not sufficient.
- (2) If statement (B) ALONE is sufficient, but statement (A) alone is not sufficient.
- (3) If BOTH statements TOGETHER are sufficient, but NEITHER statement ALONE is sufficient.
- (4) If statements (A) and (B) TOGETHER are NOT sufficient.

**Questions :**

16. If  $\frac{x+y}{z} > 0$ , is  $x < 0$  ?

- (A)  $x < y$  (B)  $z < 0$

17. If p is the perimeter of rectangle Q, what is the value of p ?

- (A) Each diagonal of rectangle Q has length 10.  
(B) The area of rectangle Q is 48.

18. In a school, 300 students study Hindi or Urdu or both. If 100 of these students do not study Hindi, how many of these students study both Hindi and Urdu ?

- (A) Of the 300 students, 200 study Hindi or both languages.  
(B) A total of 240 of the students study Urdu.

19. If m is an integer, is m odd ?

- (A)  $\frac{m}{2}$  is NOT an even integer.  
(B)  $m - 3$  is an even integer.

20. Pointing to a photograph, a woman says, "This man's son's sister is my mother-in-law". How is the woman's husband related to the man in the photograph ?

- (A) Grandson (B) Son  
(C) Son-in-law (D) Nephew

**Direction for questions 21 - 24 :**

In each of the following question, a pair of words is given. You are to study the relation existing between them and then find out from the given alternatives, the pair of words that bears the same relation between them and indicate that on the answer sheet.

**Questions :**

21. Fatigue : Resting

- (A) Overweight : Dieting  
(B) Ward : Comfortable  
(C) Sporadic : Infrequent  
(D) Elevated : Exalted

22. Triangle : Quadrilateral

- (A) Cube : Trifold  
(B) Square : Rectangle  
(C) Trident : Trapezium  
(D) Pentagon : Hexagon

23. Numismatist : Coins

- (A) Philatelist : Stamps  
(B) Jeweller : Jewels  
(C) Cartographer : Maps  
(D) Geneticist : Chromosomes





24. Textile : Mill  
 (A) Eggs : Hen  
 (B) Coal : Mine  
 (C) Food : Agriculture  
 (D) Brick : Kiln

25. Rahim moves 20 metres in East direction and then turns to his left and then moves 15 metres and then he turns to his right and moves 25 metres. After this he turns to his right and moves 15 metres. Now how far is he from his starting point?  
 (A) 0 metre (B) 40 metres  
 (C) 45 metres (D) 50 metres

Direction for questions 26 - 28 :

The numbers in each series proceed according to a certain rule. Your task is to find out the rule according to which the numbers are arranged and find out the number which can fill in the LAST blank with the '?' mark from among the suggested answers.

Questions :

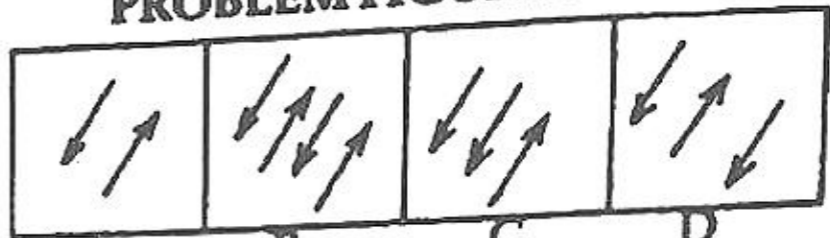
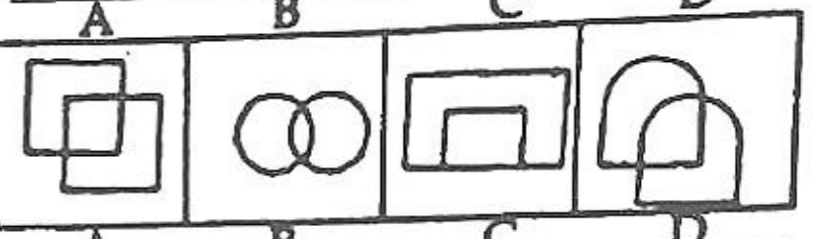
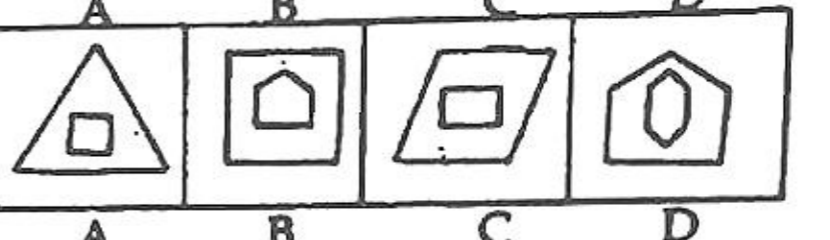
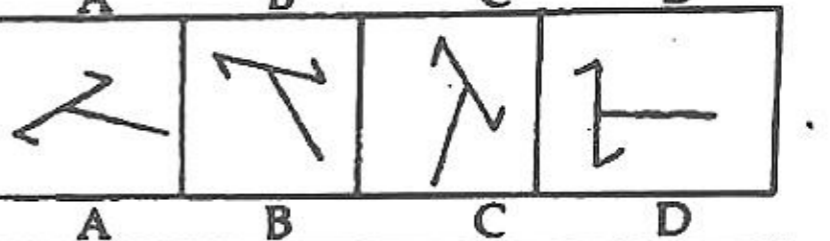
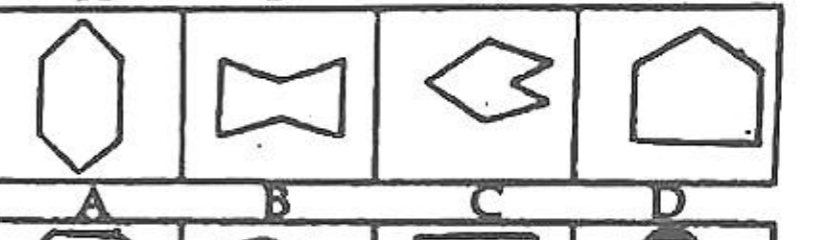
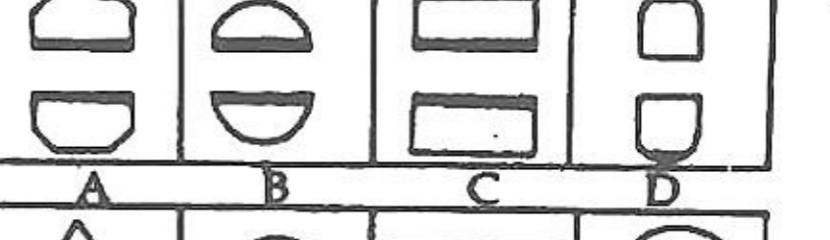
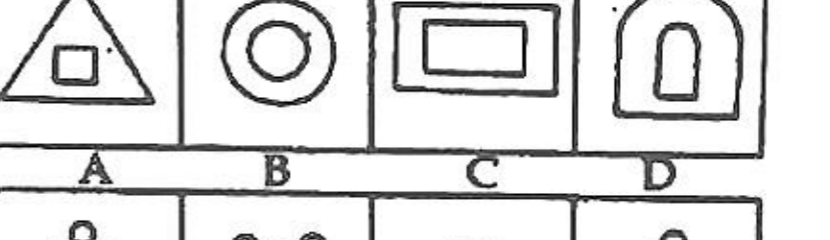
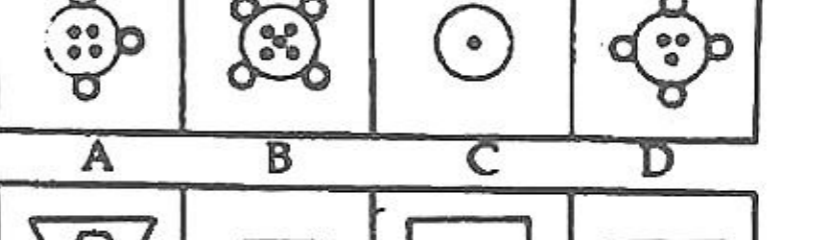

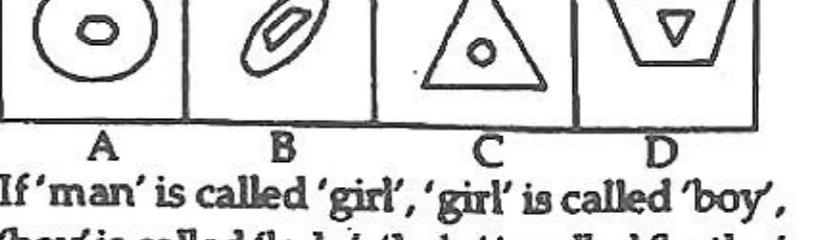
26. 301 291 282 274 ?  
 (A) 265 (B) 268  
 (C) 270 (D) 267
27. 3 2 9 4 81 16 6561 ?  
 (A) 64 (B) 243  
 (C) 256 (D) None of these
28. 4 6 10 18 34 66 ?  
 (A) 100 (B) 130  
 (C) 88 (D) 99
29. If the first day of the year (other than the leap year) was Friday, then which will be the last day of that year?  
 (A) Monday (B) Friday  
 (C) Saturday (D) Sunday

Direction for questions 30 - 39 :

In each of the following questions there are figures. Three of them are similar in some respect while one is different. Select the figure which is DIFFERENT.

Questions :

PROBLEM FIGURES

30. 
31. 
32. 
33. 
34. 
35. 
36. 
37. 
38. 
39. 
40. If 'man' is called 'girl', 'girl' is called 'boy', 'boy' is called 'lady', 'lady' is called 'butler' and 'butler' is called 'player', who will serve in restaurant?  
 (A) Player (B) Butler  
 (C) Boy (D) Lady



**Direction for questions 41 - 43 :**

All the 30 students of a class took a test on Mathematics and a test on English. The following table shows the result:

Subject	Passed	Failed
Mathematics	25	5
English	22	8

Two students failed both in Mathematics and English.

**Questions :**

41. How many students failed in only one subject ?  
(A) 28 (B) 25  
(C) 22 (D) 9
42. How many students passed in at least one subject ?  
(A) 28 (B) 25  
(C) 22 (D) 9
43. How many students passed in both the subjects ?  
(A) 28 (B) 25  
(C) 22 (D) 19
- 
44. If in a certain language, TIGER is coded as RIGET, how is 'CROWN' coded in that code ?  
(A) NROWC (B) NOWRC  
(C) RRWCO (D) NOWCR

**Direction for questions 45 - 49 :**

Read the following information to answer the given questions.

- (i) There is a group of five persons A, B, C, D and E.  
(ii) One of them is a horticulturist, one is a physicist, one is a journalist, one is an industrialist and one is an advocate.

(iii) Three of them - A, C and advocate prefer tea to coffee and two of them - B and the journalist prefer coffee to tea.

(iv) The industrialist and D and A are friend of one another but two of these prefer coffee to tea.

(v) The horticulturist is C's brother.

**Questions :**

45. Who is a horticulturist ?  
(A) A (B) B  
(C) C (D) D
46. Who is an industrialist ?  
(A) E (B) C  
(C) B (D) D
47. Which of the following groups includes a person who likes tea but is not an advocate ?  
(A) ACE (B) DE  
(C) BD (D) BCD
48. Who is a physicist ?  
(A) A (B) C  
(C) D (D) E
49. Which of the statements above is superfluous in order to answer the above questions ?  
(A) Nil (B) (iii)  
(C) (ii) (D) (v)
- 
50. If  $BAT = 40$ ,  $AT = 20$  then  $CAT$  will be equal to ?  
(A) 70 (B) 50  
(C) 60 (D) 30



## PHYSICS

Question : 50

51. The rate of change of current of  $10 \text{ A s}^{-1}$  in a coil produces an emf of 5 V. Then the self-inductance of the coil in henry is :
- (A) 0.5 (B) 0.25  
(C) 1.0 (D) 1.25
52. The phase difference between the alternating current and emf is  $\pi/2$ . Which of the following CANNOT be the constituent of the circuit ?
- (A) L, C (B) L alone  
(C) C alone (D) R, L
53. An electric motor operating on 15 V supply draws a current of 5 A and yields mechanical powers of 60 W. The energy lost as heat in one hour (in kJ) is :
- (A) 0.54 (B) 5.4  
(C) 54 (D) 540
54. The ratio of amplitude of magnetic field to the amplitude of electric field for an electromagnetic wave propagating in vacuum is equal to :
- (A) the speed of light in vacuum  
(B) reciprocal of speed of light in vacuum  
(C) the ratio of magnetic permeability to the electric susceptibility of vacuum  
(D) unity
55. The ratio of the speed of an object to the speed of its real image of magnification  $m$  in the case of a convex mirror is :
- (A)  $-1/m^2$  (B)  $m^2$   
(C)  $-m$  (D)  $1/m$
56. An air bubble is contained inside water. It behaves as a :
- (A) concave lens  
(B) convex lens  
(C) neither concave nor convex  
(D) None of these
57. An eye specialist prescribes spectacles having a combination of a convex lens of focal length 40 cm in contact with a concave lens of focal length 25 cm. The power of this lens combination will be :
- (A) +1.5 D (B) -1.5 D  
(C) +6.67 (D) -6/67
58. Two monochromatic light waves of amplitudes A and 2A interfering at a point have a phase difference of  $60^\circ$ . The intensity at that point will be proportional to :
- (A)  $3A^2$  (B)  $5A^2$   
(C)  $7A^2$  (D)  $9A^2$
59. Photoelectric emission occurs only when the incident light has more than a certain minimum :
- (A) power (B) wavelength  
(C) intensity (D) frequency
60. If the kinetic energy of a free electron doubles, its de Broglie wavelength changes by the factor
- (A)  $1/\sqrt{2}$  (B)  $\sqrt{2}$   
(C)  $1/2$  (D) 2
61. The decimal equivalent of the binary number  $(11010.101)_2$  is :
- (A) 9.625 (B) 25.265  
(C) 26.625 (D) 26.265
62. Application of a forward bias to a p-n junction :
- (A) widens the depletion zone  
(B) increases the potential difference across the depletion zone  
(C) increases the number of donors on the n-side  
(D) increases the electric field in the depletion zone
63. If force (F), velocity (V) and time (T) are taken as fundamental units, then the dimensions of mass are :
- (A)  $FV^{-1}T^{-1}$  (B)  $FVT^{-1}$   
(C)  $FV^{-1}T$  (D)  $FVT^2$





64. If the error in measurement of radius of a sphere is 2%, then the error in the determination of volume of the sphere will be :
- (A) 2% (B) 4%  
~~(C) 6%~~ (D) 8%
65. A particle is travelling along a straight line OX. The distance  $x$  (in meters) of the particle from O at a time  $t$  is given by  $x = 37 + 27t - t^3$ , where  $t$  is in seconds. The distance of the particle from O when it comes to rest is :
- (A) 81 m (B) ~~91 m~~  
 (C) 101 m (D) 111 m
66. Value of ratio between cross product and dot product of two vectors is  $1/\sqrt{3}$ . The angle between two vectors is :
- (A) ~~30°~~ (B) 45°  
 (C) 60° (D) 120°
67. A body of mass  $M$  hits normally a rigid wall with velocity  $V$  and bounces back with the same velocity. The impulse experienced by the body is :
- (A)  $MV$  (B)  $1.5MV$   
~~(C)  $2MV$~~  (D) Zero
68. For an object sliding on a plane, the force of friction is less if the plane is inclined, instead of being horizontal
- (A) because, effective mass decreases  
 (B) because, normal force decreases  
~~(C) because, co-efficient of friction decreases~~  
 (D) for an angle of inclination  $\theta$ , friction is inversely proportional to  $\tan \theta$
69. If a body travels along a circular path with uniform speed then its acceleration
- (A) is zero  
 (B) acts along its circumference  
 (C) acts along its tangent  
~~(D) acts along its radius~~
70. A force is acting on a mass of 6 kg. Displacement  $x$  of the mass is related with time  $t$  as  $x = t^2/4$  meter. Work done by the force in 2 sec. is :
- ~~(A) 3 J~~ (B) 6 J  
 (C) 9 J (D) 12 J
- ~~71.~~ A particle is projected from the ground with kinetic energy  $E$  at an angle of  $60^\circ$  with the horizontal. Its kinetic energy at the highest point of its motion will be :
- (A)  $E/\sqrt{2}$  (B)  $E/2$   
 (C)  $E/4$  (D)  $E/8$
72. Power required to raise a mass of 120 kg vertically upwards at a velocity of  $4.5 \text{ m.s}^{-1}$  is :
- (A) 5 kW (B) ~~5.3 kW~~  
 (C) 8 kW (D) 11.2 kW
73. A body falling vertically downwards under gravity breaks in two parts of unequal masses. The centre of mass of the two parts taken together shifts horizontally towards
- (A) lighter piece  
 (B) heavier piece  
~~(C) depends on the vertical velocity at the time of breaking~~  
 (D) does not shift horizontally
74. Angular momentum of a moving body remains constant if
- (A) a pressure acts on the body  
 (B) an external force acts on the body  
 (C) an external torque acts on the body  
~~(D) no external torque acts on the body~~
75. Moment of inertia of circular wire of mass  $m$  and radius  $r$  about its diameter is
- ~~(A)  $1/2mr^2$~~  (B)  $1/4 mr^2$   
 (C)  $mr^2$  (D)  $2mr^2$





76. If the diurnal motion of the earth ceases all on a sudden, then the value of the acceleration due to gravity of a body at the equator will  
 (A) remains same (B) be zero  
 (C) increase (D) decrease
77. Two satellites of masses  $m_1$  and  $m_2$  ( $m_1 > m_2$ ) are revolving around the earth in orbits of radii  $r_1$  and  $r_2$  ( $r_1 > r_2$ ) with velocities  $v_1$  and  $v_2$  respectively. In this case  
 (A)  $v_1 = v_2$  (B)  $v_1 < v_2$   
 (C)  $v_1 > v_2$  (D)  $v_1/r_1 = v_2/r_2$
78. A wire of initial length  $L$  and area of cross-section  $A$  has Young's modulus  $Y$  of its material. The wire is stretched by a stress  $S$  within its elastic limit. The stored energy density in the wire will be  
 (A)  $S/2Y$  (B)  $2Y/S^2$   
 (C)  $S^2/2Y$  (D)  $S^2/Y$
79. Which of the following works on Pascal's law?  
 (A) Sprayer  
 (B) Hydraulic lift  
 (C) Barometer (D) Venturimeter
80. Surface energy of a water drop of radius  $r$  will be directly proportional to  
 (A)  $r$  (B)  $r^2$   
 (C)  $r^3$  (D)  $1/r$
81. A spherical ball is falling with a uniform velocity  $v$  through a viscous medium of coefficient of viscosity  $\eta$ . If the viscous force acting on the spherical ball is  $F$  then  
 (A)  $F \propto \eta$  and  $F \propto 1/v$   
 (B)  $F \propto \eta$  and  $F \propto v$   
 (C)  $F \propto 1/\eta$  and  $F \propto 1/v$   
 (D)  $F \propto 1/\eta$  and  $F \propto v$
82. Apparent weight of a body immersed in water at  $20^\circ\text{C}$  is  $W_1$ . When temperature is increased to  $40^\circ\text{C}$ , the apparent weight becomes  $W_2$ . In this case  
 (A) for different solids  $W_2$  may be greater than or less than  $W_1$   
 (B)  $W_2$  is always equal to  $W_1$   
 (C)  $W_2$  is always less than  $W_1$   
 (D)  $W_2$  is always greater than  $W_1$
83. An ideal gas is expanding such that  $pT^2 = \text{constant}$ . The coefficient of volume expansion of the gas is  
 (A)  $1/T$  (B)  $2/T$   
 (C)  $3/T$  (D)  $4/T$
84. During boiling water at  $100^\circ\text{C}$ , what will be its specific heat?  
 (A) zero (B) 0.5  
 (C) 1 (D) infinite
85. If the temperature of a black body raises from  $T$  to  $2T$ , how many times will its rate of radiation be?  
 (A) 16 (B) 8  
 (C) 4 (D) 2
86. In a given process of an ideal gas,  $dW = 0$  and  $dQ < 0$ . Then for the gas  
 (A) the temperature will decrease  
 (B) the volume will increase  
 (C) the pressure will remain constant  
 (D) the temperature will increase
87. Even Carnot engine CANNOT give 100% efficiency, because we CANNOT  
 (A) eliminate friction  
 (B) prevent radiation  
 (C) reach absolute zero temperature  
 (D) find ideal sources
88. 3 mol of a monatomic gas ( $\nu = 5/3$ ) is mixed with 1 mol of a diatomic gas ( $\nu = 7/3$ ). The value of  $\nu$  for the mixture will be  
 (A)  $9/11$  (B)  $11/7$   
 (C)  $12/7$  (D)  $15/7$





89. Time period of a simple pendulum on the surface of the earth is  $T_1$  and at a height  $R$  above the surface of the earth is  $T_2$ , here  $R$  is the radius of the earth. The ratio  $T_1/T_2$  :
- (A) 1 (B)  $\sqrt{2}$   
 (C) 4 (D) 2
90. When a force  $F_1$  acts on a particle, frequency is 6 Hz and when a force  $F_2$  acts, frequency is 8 Hz. Now if both the forces act simultaneously in same direction, then its frequency becomes
- (A) 20 Hz (B) 14 Hz  
 (C) 10 Hz (D) 2 Hz
91. When a fundamental tone is produced from a pipe of length  $l$ , open at both ends, the wavelength of the stationary wave is
- (A)  $l$  (B)  $2l$   
 (C)  $l/2$  (D)  $4l$
92. A source of sound with frequency 256 Hz is moving with a velocity  $v$  towards a wall and an observer is stationary between the source and wall. When the observer is between the source and the wall, he will
- (A) hear beats (B) hear no beats  
 (C) not get any sound  
 (D) get the sound of same frequency
93. If a charge  $q$  is placed at the centre of the line joining two equal charges  $Q$  such that the system is in equilibrium, then the value of  $q$  is
- (A)  $Q/2$  (B)  $-Q/2$   
 (C)  $-Q/4$  (D)  $Q/4$
94. If a linear isotropic dielectric is placed in an electric field of strength  $E$ , then the polarisation  $P$  is
- (A) independent of  $E$   
 (B) inversely proportional to  $E$   
 (C) directly proportional to  $\sqrt{E}$   
 (D) directly proportional to  $E$
95. Three resistances  $P, Q, R$  each of 2 ohm and an unknown resistance  $S$  form the four arms of a Wheatstone's bridge circuit. When a resistance of 6 ohm is connected in parallel to  $S$  the bridge gets balanced. What is the value of  $S$  ?
- (A)  $2\Omega$  (B)  $3\Omega$   
 (C)  $6\Omega$  (D)  $1\Omega$
96. A heater coil is cut into two equal parts and only one part is now used in the heater. The heat generated will now be
- (A) one fourth (B) halved  
 (C) doubled (D) four times
97. An electric bulb marked as 50 W-200 V is connected across a 100 V supply. The present power of the bulb is
- (A) 37.5 W (B) 25 W  
 (C) 12.5 W (D) 10 W
98. A circular coil carrying a certain current produces a magnetic field  $B_0$  at its centre. The coil is now rewound so as to have 3 turns and the same current is passed through it. The new magnetic field at the centre is
- (A)  $B_0/9$  (B)  $9B_0$   
 (C)  $B_0/3$  (D)  $3B_0$
99. A straight wire of length 2 m carries a current of 10 A. If this wire is placed in a uniform magnetic field of 0.15 T making an angle of  $45^\circ$  with the magnetic field, the applied force on the wire will be
- (A) 1.5 N (B)  $3\sqrt{2}$  N  
 (C) 3 N (D)  $3/\sqrt{2}$
100. Nickel shows ferromagnetic property at room temperature. If the temperature is increased beyond Curie temperature, then it will show
- (A) anti-ferromagnetism  
 (B) paramagnetism  
 (C) diamagnetism  
 (D) no magnetic property





**CHEMISTRY**

Question : 50

101. At 250°C, the correct order of molar ionic conductances of the ions  $H^+$ ,  $Li^+$ ,  $Na^+$  and  $K^+$  in infinite dilute aqueous solution is  
(A)  $H^+ < Li^+ < Na^+ < K^+$   
(B)  $K^+ < Na^+ < Li^+ < H^+$   
(C)  $Li^+ < Na^+ < K^+ < H^+$   
(D)  $Li^+ < K^+ < H^+ < Na^+$
102. The activation energy of a reaction depends on  
(A) temperature  
(B) initial concentration of the reactant  
(C) effective collisions among the reactant molecules  
(D) nature of the reactants
103. The bottle of liquor ammonia is cooled before opening the cork because it  
(A) is a mild explosive  
(B) is a corrosive liquid  
(C) is harmful to lung  
(D) exerts high vapour pressure
104. Which of the following substances form a colloidal solution in water?  
(A) Glucose (B) Urea  
(C)  $BaSO_4$  (D) Starch
105. Adsorption of a gas on a solid surface is an exothermic process, because  
(A) change in free energy of the system increases  
(B) enthalpy of the system increases  
(C) entropy of the system increases  
(D) enthalpy of the system decreases
106. In the manufacture of steel, the process in which  $O_2$  is used instead of air is  
(A) Open-hearth process  
(B) Acidic Bessemer's process  
(C) Alkaline Bessemer's process  
(D) LD process
107. The ore that does NOT contain aluminium is  
(A) Fluorspar (B) Feldspar  
(C) Cryolite (D) Mica
108. Which of the following nitrogen oxides is ionic?  
(A) Nitrogen trioxide  
(B) Nitrogen pentoxide  
(C) Dinitrogen tetroxide  
(D) Nitric oxide
109. Which one of the following is used as the photosensitive substance in xerox machines  
(A) Hg (B) Black P  
(C) Se (D) Te
110.  $Fe^{2+}$  can be differentiated from  $Fe^{3+}$  with the help of  
(A)  $BaCl_2$  (B)  $AgNO_3$   
(C)  $NH_4SCN$  (D) None of these
111. The salt of the d-block element that is used as a catalyst in the dissociation of the bleaching powder is  
(A) Ni (B) CO  
(C) V (D) Cr
112. The reagent used for identifying Nickel ion is  
(A) Potassium ferrocyanide  
(B) Phenolphthalein  
(C) Dimethylglyoxime  
(D) EDTA
113. When 800g of a 40% solution by weight was cooled, 100 g of solute was precipitated. The percentage composition of the remaining solution is  
(A) 20.0% (B) 25.0%  
(C) 31.4% (D) 50.0%
114. A sample of  $Na_2CO_3 \cdot H_2O$  weighing 0.62 g is added to 100 ml of 0.1 N  $H_2SO_4$ . The resulting solution will be  
(A) Basic (B) Neutral  
(C) Acidic (D) Amphoteric
115. An anion  $X^{3-}$  has 36 electrons and 45 neutrons. What is the mass number of the element X?  
(A) 81 (B) 84  
(C) 78 (D) 88





116. If two particles are associated with same kinetic energy, then the de-Broglie's wavelength ( $\lambda$ ) of these particles is  
 (A) directly proportional to the velocity  
 (B) inversely proportional to the velocity  
 (C) independent of mass and velocity  
 (D) cannot be predicted
117. The increasing order of the first ionisation enthalpies of the elements B, P, S and F is  
 (A)  $F < S < P < B$  (B)  $P < S < B < F$   
 (C)  $B < P < S < F$  (D)  $B < S < P < F$
118. In the relation,  $\text{Electronegativity} = \frac{0.359 Z_{\text{eff}}}{r^2} + 0.744$ ,  $r$  is  
 (A) Metallic radius  
 (B) Ionic radius  
 (C) van der Waals radius  
 (D) Covalent radius
119. The species in which the central atom uses  $sp^2$  hybrid orbitals in its bonding is  
 (A)  $\text{NH}_3$  (B)  $\text{PH}_3$   
 (C)  $\text{CH}_3^+$  (D)  $\text{SbH}_3$
120. The molecule with the highest dipole moment is  
 (A)  $\text{CH}_2\text{Cl}_2$  (B)  $\text{CH}_3\text{Cl}$   
 (C)  $\text{CHCl}_3$  (D)  $\text{CCl}_4$
121. The dimension of coefficient of viscosity  
 (A)  $\text{MLT}$  (B)  $\text{ML}^{-1}\text{T}^{-1}$   
 (C)  $\text{MLT}^{-1}$  (D)  $\text{MLT}^{-2}$
122. At STP,  $\text{O}_2$  gas present in a flask was replaced by  $\text{SO}_2$  under similar conditions. The mass of  $\text{SO}_2$  present in the flask will be  
 (A) half that of  $\text{O}_2$   
 (B) equal to that of  $\text{O}_2$   
 (C) twice that of  $\text{O}_2$   
 (D) one-third of  $\text{O}_2$
123. In a reversible process, if the changes in entropy of the system and its surroundings are  $\Delta S_1$  &  $\Delta S_2$  respectively, then  
 (A)  $\Delta S_1 + \Delta S_2 > 0$  (B)  $\Delta S_1 + \Delta S_2 < 0$   
 (C)  $\Delta S_1 + \Delta S_2 = 0$  (D)  $\Delta S_1 + \Delta S_2 \geq 0$
124. The volume of a gas is reduced to half from its original volume. The specific heat will  
 (A) double (B) remain constant  
 (C) reduce to half  
 (D) increase four times
125. The reaction,  $\text{A(g)} + 2\text{B(g)} \rightleftharpoons 2\text{C(g)} + \text{D(g)}$  was studied using an initial concentration of B which was 1.5 times that of A. The equilibrium concentration of A and C were found to be equal. So,  $K_c$  for the equilibrium is  
 (A) 0.32 (B) 2.73  
 (C) 4.0 (D) 8.17
126. A mixture containing  $\text{N}_2$  and  $\text{H}_2$  in a mole ratio 1 : 3 is allowed to attain equilibrium when 50% of the mixture has reacted. If  $P$  is the pressure at equilibrium, then the partial pressure of  $\text{NH}_3$  formed is  
 (A)  $P/2$  (B)  $P/3$   
 (C)  $P/5$  (D)  $P/9$
127. Oxidation number of P in pyrophosphoric acid is  
 (A) +1 (B) +3  
 (C) +4 (D) +5
128. The amount of  $\text{H}_2\text{O}_2$  required for decolourising 1 mol of  $\text{KMnO}_4$  in an acid solution is  
 (A) 1.5 mol (B) 2.0 mol  
 (C) 2.5 mol (D) 3.0 mol
129. The process by which hydrogen is prepared by the reaction of silicon, iron alloy and  $\text{NaOH}$ , is  
 (A) Haber's process (B) Silicon process  
 (C) Wood process (D) Bosch process
130. Which of the followings does NOT get reduced by  $\text{H}_2$  in its aqueous solution?  
 (A)  $\text{Cu}^{2+}$  (B)  $\text{Fe}^{3+}$   
 (C)  $\text{Zn}^{2+}$  (D)  $\text{Ag}^+$
131. The compound which is used to extinguish fire caused by combustion of alkali metals is  
 (A)  $\text{CCl}_4$  (B) Sand  
 (C) Water (D) Kerosene
132. The compound whose aqueous solution is called 'baryta water' is  
 (A)  $\text{BaSO}_4$  (B)  $\text{BaO}$   
 (C)  $\text{BaCO}_3$  (D)  $\text{Ba(OH)}_2$





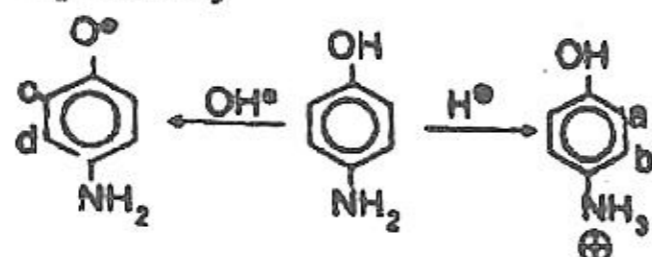
133. The optically active alkane of lowest molecular mass which is also chiral is  
 (A) 3-methylhexane  
 (B) 2,3-dimethylpentane  
 (C) 2-methylhexane  
 (D) 2, 5-dimethylhexane

134. Bond lengths of C - H, C - O, C - C and C = C follow the sequence  
 (A) C - H < C - O < C - C < C = C  
 (B) C - H < C = C < C - O < C - C  
 (C) C - C < C = C < C - O < C - H  
 (D) C - O < C - H < C - C < C = C

135. Nitrobenzene is prepared from benzene by using conc. HNO<sub>3</sub> and conc. H<sub>2</sub>SO<sub>4</sub>. In the nitrating mixture, nitric acid acts as a/an

- (A) Base (B) Acid  
 (C) Reducing agent (D) Catalyst

136. In strong acidic and alkaline medium, *p*-aminophenol exists in (X) and (Y) forms respectively



Thus, in acidic and alkaline medium, electrophilic substitution occurs at

- (A) a, c (B) a, d  
 (C) b, c (D) b, d

137. Incomplete combustion of gasoline produces

- (A) CO<sub>2</sub> (B) CO  
 (C) SO<sub>2</sub> (D) NO<sub>2</sub>

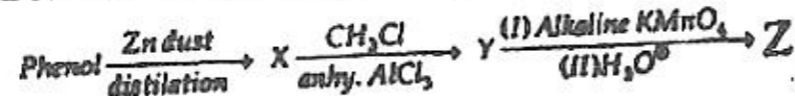
138. Cause of byssinosis diseases

- (A) fly-ash (B) cement particles  
 (C) cotton fibre (D) lead particles

139. Which one is NOT favourable for S<sub>N</sub>1 reaction

- (A) Polar solvent  
 (B) Strong nucleophile  
 (C) Low concentration  
 (D) 3° alkyl halide

140. Consider the following reaction :



The product Z is

- (A) Benzaldehyde (B) Benzene  
 (C) Benzoic acid (D) Toluene

141. Which converts carboxylic acids directly into alcohols ?

- (A) LiAlH<sub>4</sub> (B) Na + C<sub>2</sub>H<sub>5</sub>OH  
 (C) NaBH<sub>4</sub> (D) All of these

142. In the reaction of acetaldehyde with aniline, the product formed is

- (A) Schiff's base (B) Carbylamine  
 (C) Imine (D) None of these

143. Complete hydrolysis of cellulose yields

- (A) D-fructose (B) D-ribose  
 (C) D-glucose (D) L-glucose

144. Monomers of Buna-S are

- (A) Styrene and Butadiene  
 (B) Butadiene  
 (C) Isoprene and Butadiene  
 (D) Vinyl chloride and Sulphur

145. Chemical name of aspirin is

- (A) Methyl Benzoate  
 (B) Ethyl Salicylate  
 (C) Acetylsalicylic acid  
 (D) Hydroxybenzoic acid

146. Which of the following crystal systems does NOT have body-centered lattice ?

- (A) Orthorhombic (B) Tetragonal  
 (C) Monoclinic (D) Cubic

147. NaCl has face-centered unit cell. In its crystal, the number of Cl<sup>-</sup> ions present in contact with a Na<sup>+</sup> ion is

- (A) 4 (B) 6  
 (C) 8 (D) 10

148. Which of the following concentration units does NOT depend on temperature ?

- (A) Molarity (B) Normality  
 (C) Mole fraction (D) Formality

149. At a given temperature, which one of the following solutions would have the highest vapour pressure ?

- (A) 0.1 m glucose solution  
 (B) 0.1 m NaCl solution  
 (C) 0.1 m CaCl<sub>2</sub> solution  
 (D) 0.1 m Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> solution

150. Which one of the following does NOT give precipitate on reaction with lead acetate ?

- (A) HI (B) HBr  
 (C) HCl (D) HF





# MATHEMATICS

## Question : 50

151. The largest interval for which  $x^{12} - x^9 + x^4 - x + 1 > 0$ . is  
 (A)  $-4 < x \leq 0$  (B)  $0 < x < 1$   
 (C)  $-100 < x < 100$  (D)  $0 < x < \infty$
152. The smallest positive integer  $n$  for which  $n! < \left(\frac{n+1}{2}\right)^n$  holds is  
 (A) 1 (B) 2  
 (C) 3 (D) 4
153. The number of integral solutions of  $x^2 + y^2 = x^2y^2$  is  
 (A) 0 (B) 1  
 (C) infinite (D) None of these
154. How many ways are there to arrange the letters in the word GARDEN with the vowels in alphabetical order?  
 (A) 120 (B) 240  
 (C) 360 (D) 480
155. The sum of the series  $\frac{1}{1.2} - \frac{1}{2.3} + \frac{1}{3.4} \dots$  upto  $\infty$  is equal to  
 (A)  $2 \log_e 2$  (B)  $\log_e 2 - 1$   
 (C)  $\log_e 2$  (D)  $\log_e \left(\frac{4}{e}\right)$
156. Let  $a_1, a_2, a_3, \dots$  cannot be terms of an A.P.  
 If  $\frac{a_1 + a_2 + \dots + a_p}{a_1 + a_2 + \dots + a_q} = \frac{p^2}{q^2}, p \neq q$   
 then  $\frac{a_6}{a_{21}}$  equals  
 (A)  $\frac{7}{2}$  (B)  $\frac{2}{7}$   
 (C)  $\frac{11}{41}$  (D)  $\frac{41}{11}$
157. If the slope of the line joining the points  $A(x, 2)$  and  $B(6, -8)$  is  $-\frac{5}{4}$ , then  $x = ?$   
 (A) -2 (B) 2  
 (C) -3 (D) 3
158. The equation of the perpendicular bisector of the line joining the points  $A(2, 3)$  and  $B(6, -9)$  is  
 (A)  $x + 2y - 6 = 0$  (B)  $x - 2y - 6 = 0$   
 (C)  $x + 2y + 6 = 0$  (D)  $x - 2y + 6 = 0$
159. If  $A(-1, 3)$  and  $B(\alpha, \beta)$  be the extremities of the diameter of the circle  $x^2 + y^2 - 6x + 5y - 7 = 0$ , then  
 (A)  $\alpha = -7, \beta = 8$  (B)  $\alpha = 7, \beta = -8$   
 (C)  $\alpha = -6, \beta = 7$  (D)  $\alpha = 6, \beta = -7$
160. If the parabola  $y^2 = 4ax$  passes through the point  $P(3, 2)$ , then the length of its latus rectum is  
 (A)  $\frac{1}{3}$  (B)  $\frac{2}{3}$   
 (C)  $\frac{4}{3}$  (D) 4
161. One focus of hyperbola is at  $(0, 4)$  and the length of its transverse axis is 6. The equation of the hyperbola is  
 (A)  $\frac{x^2}{7} - \frac{y^2}{9} = 1$  (B)  $\frac{y^2}{9} - \frac{x^2}{7} = 1$   
 (C)  $\frac{y^2}{4} - \frac{x^2}{9} = 1$  (D) None of these
162. The foci of an ellipse are  $(0, \pm 6)$  and the length of its minor axis is 16. The equation of the ellipse is  
 (A)  $\frac{x^2}{16} + \frac{y^2}{36} = 1$  (B)  $\frac{x^2}{36} + \frac{y^2}{64} = 1$   
 (C)  $\frac{x^2}{100} + \frac{y^2}{64} = 1$  (D)  $\frac{x^2}{64} + \frac{y^2}{100} = 1$
163. State which of the following is total number of relations from set  $A = \{1, 2, 3, 4\}$  to set  $B = \{d, e\}$  is -  
 (A)  $2^4$  (B)  $2^6$   
 (C)  $2^8$  (D)  $2^{15}$
164. Let  $A = \{a, b, c, d\}$  and  $f: A \rightarrow A$  be defined by,  $f(a) = d, f(b) = a, f(c) = b$  and  $f(d) = c$ . State which of the following is equal to  $f^4(b)$ ?  
 (A)  $\{a\}$  (B)  $\{b\}$   
 (C)  $\{c\}$  (D)  $\{d\}$





165. If the binary operation on  $Z$  is defined by  $a * b = a^2 - b^2 + ab + 4$ , then the value of  $(2 * 3) * 4$  is  
 (A) 233 (B) 33  
 (C) 55 (D) -55
166. If  $\sin^{-1} x - \cos^{-1} x = \frac{\pi}{6}$ , state which of the following is the value of  $x$ ?  
 (A) 1 (B)  $\frac{1}{2}$   
 (C)  $\frac{1}{\sqrt{2}}$  (D)  $\frac{\sqrt{3}}{2}$
167. Given the LPP, min.  $Z = 3x - y$ , subject to the constraints  
 $2x + 3y \geq 1$  and  $x, y \geq 0$   
 The optimum solution of the LPP is  
 (A)  $x = 0, y = \frac{1}{2}$  (B)  $x = 0, y = \frac{1}{3}$   
 (C)  $x = \frac{1}{3}, y = 0$  (D)  $x = \frac{1}{2}, y = 0$
168. In a linear programming problem, the equation  $2x + 3y = 12$  in two unknowns has number of solutions equal to  
 (A) maximum value of and minimum value of  
 (B) a particular value of and  
 (C) infinite (D) None of these
169. Let  $A$  be a square matrix of order  $3 \times 3$ , then  $|KA|$  is equal to  
 (A)  $K|A|$  (B)  $K^2|A|$   
 (C)  $K^3|A|$  (D)  $3K|A|$
170. The system of equations  
 $\alpha x + y + z = \alpha - 1$   
 $x + \alpha y + z = \alpha - 1$   
 $x + y + \alpha z = \alpha - 1$   
 has no solution, if  $\alpha$  is  
 (A) 1 (B) not -2  
 (C) either -2 or 1 (D) -2
171. Matrix  $A = \begin{bmatrix} 1 & 0 & -k \\ 2 & 1 & 3 \\ k & 0 & 1 \end{bmatrix}$  is invertible for  
 (A)  $k = 1$  (B)  $k = -1$   
 (C) all real  $k$  (D) None of these
172. If matrix  $A = \begin{bmatrix} 3 & 2 & 4 \\ 1 & 2 & -1 \\ 0 & 1 & 1 \end{bmatrix}$  and  $A^{-1} = \frac{1}{k} \text{adj } A$ , then  $k$  is  
 (A) 7 (B) -7  
 (C)  $\frac{1}{7}$  (D) 11
173. The mean and the variance of a binomial distribution are 4 and 2 respectively. Then the probability of 2 successes is  
 (A)  $\frac{37}{256}$  (B)  $\frac{219}{256}$   
 (C)  $\frac{128}{256}$  (D)  $\frac{28}{256}$
174. If  $P(A \cup B) = 0.8$  and  $P(A \cap B) = 0.3$ , then  $P(A') + P(B')$  equals to  
 (A) 0.9 (B) 0.7  
 (C) 0.5 (D) 0.3
175. The value of  $f$  at  $x = 0$  so that function  $f(x) = \frac{2^x - 2^{-x}}{x}, x \neq 0$ , is continuous at  $x = 0$ , is  
 (A) 0 (B)  $\log 2$   
 (C)  $\log 4$  (D)  $e^4$
176. The rate of change of the function  $y = f(x)$  w.r.t  $x$  at the point  $x$  is  
 (A)  $\frac{1}{2}f'(x)$  (B)  $2f'(x)$   
 (C)  $\frac{f'(x)}{f(x)}$  (D) None of these
177. The slope of the tangent to the ellipse  $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$  at the point  $(a \cos \theta, b \sin \theta)$  is  
 (A)  $\frac{b}{a} \tan \theta$  (B)  $\frac{b}{a} \cot \theta$   
 (C)  $-\frac{b}{a} \tan \theta$  (D)  $-\frac{b}{a} \cot \theta$
178. A function  $f(x)$  is defined in  $a < x < b$  and  $a \leq x_1 \leq x_2 < b$ ; then  $f(x)$  is strictly monotonic decreasing in  $a \leq x \leq b$  when  
 (A)  $f(x_2) > f(x_1)$  when  $x_2 > x_1$   
 (B)  $f(x_2) < f(x_1)$  when  $x_2 > x_1$   
 (C)  $f(x_2) > f(x_1)$  when  $x_2 < x_1$   
 (D)  $f(x_2) < f(x_1)$  when  $x_2 < x_1$





179. If  $0 \leq x \leq 2\pi$ , the function  $f(x) = \sin x$  is minimum at  
 (A)  $x = \frac{3\pi}{2}$  (B)  $x = \pi$   
 (C)  $\frac{3\pi}{4}$  (D)  $x = 2\pi$
180. The value of  $\int (\operatorname{cosec} 2x \cot 2x) dx$  is  
 (A)  $-\frac{\cot 2x}{2} + c$  (B)  $2 \cot 2x + c$   
 (C)  $-2 \operatorname{cosec} 2x + c$  (D)  $-\frac{\operatorname{cosec} 2x}{2} + c$
181. If  $\int_0^{\pi} x f(\sin x) dx = A \int_0^{\frac{\pi}{2}} f(\sin x) dx$ , then A is equal to  
 (A) 0 (B)  $\pi$   
 (C)  $\frac{\pi}{4}$  (D)  $2\pi$
182. The order and degree of the differential equation  $\left(1 + 3 \frac{dy}{dx}\right)^{\frac{2}{3}} = 4 \frac{d^3y}{dx^3}$  are  
 (A)  $1, \frac{2}{3}$  (B) 3, 1  
 (C) 3, 3 (D) 1, 2
183. The solution of the differential equation  $y dx + (x + x^2y) dy = 0$  is  
 (A)  $-\frac{1}{xy} = c$  (B)  $-\frac{1}{xy} + \log y = c$   
 (C)  $\frac{1}{xy} + \log y = c$  (D)  $\log y = cx$
184. If  $\int e^{ax} \sin bx dx = \frac{e^{ax}}{\sqrt{a^2 + b^2}} \sin(bx + m) + c$ , then the value of m is  
 (A)  $-\tan^{-1} \frac{b}{a}$  (B)  $\tan^{-1} \frac{b}{a}$   
 (C)  $\tan^{-1} \frac{a}{b}$  (D)  $-\tan^{-1} \frac{a}{b}$
185. If the position vectors of the points P and Q are  $2\hat{i} + \hat{k}$  and  $-3\hat{i} - 4\hat{j} - 5\hat{k}$  respectively, then vector  $\overline{QP}$  is  
 (A)  $5\hat{i} + 4\hat{j} + 5\hat{k}$  (B)  $5\hat{i} + 4\hat{j} + 6\hat{k}$   
 (C)  $5\hat{i} - 4\hat{j} + 4\hat{k}$  (D)  $-\hat{i} - 4\hat{j} - 4\hat{k}$
186. If a line whose direction ratios are proportional to 0, 1, -1 then the inclination of the line with z-axis is  
 (A)  $\frac{\pi}{2}$  (B)  $\pi$   
 (C)  $\frac{3\pi}{2}$  (D)  $\frac{3\pi}{4}$
187. If the line  $\frac{x-x_1}{a} = \frac{y-y_1}{b} = \frac{z-z_1}{c}$  is parallel to z-axis then  
 (A)  $a = c = 0$  and  $b \neq 0$   
 (B)  $a = b = 0$  and  $c \neq 0$   
 (C)  $b = c = 0$  and  $a \neq 0$   
 (D)  $a = b = c = 0$
188. The lines  $\frac{x}{1} = \frac{y}{2} = \frac{z}{3}$  and  $\frac{x-1}{-2} = \frac{y-2}{-4} = \frac{z-3}{6}$  are  
 (A) coincident (B) skew  
 (C) intersecting (D) parallel
189. The intercept made by the plane  $\vec{r} \cdot \vec{n} = q$  on the x-axis is  
 (A)  $\frac{q}{\hat{i} \cdot \vec{n}}$  (B)  $\frac{\hat{i} \cdot \vec{n}}{q}$   
 (C)  $-\frac{\hat{i} \cdot \vec{n}}{q}$  (D)  $\frac{q}{|\hat{n}|}$
190. Value of  $\lambda$  such that the line  $\frac{x-1}{2} = \frac{y-1}{3} = \frac{z-1}{\lambda}$  is perpendicular to normal to the plane  $\vec{r} \cdot (2\hat{i} + 3\hat{j} + 4\hat{k}) = 0$  is  
 (A)  $-\frac{13}{4}$  (B)  $-\frac{17}{4}$   
 (C) 4 (D)  $-\frac{11}{4}$
191. In the expansion of  $(1+x)^n$ , the binomial co-efficients of three consecutive terms are respectively 220, 495 and 792. The value of n is  
 (A) 10 (B) 11  
 (C) 12 (D) 13





192. The function of  $f(x) = \log(x + \sqrt{x^2 + 1})$  is

- (A) an even function  
 (B) an odd function  
 (C) a periodic function  
 (D) neither an even nor an odd function

193. If  $A = \{x : x = 4n + 1, 2 \leq n \leq 5\}$ , then the number of subsets of A is

- (A) 16 (B) 15  
 (C) 4 (D) None of these

194. If  $f : \mathbb{R} \rightarrow \mathbb{R}$  satisfies  $f(x + y) = f(x) + f(y)$  for all  $x, y \in \mathbb{R}$  and  $f(1) = 7$ ,

then  $\sum_{r=1}^n f(r)$  is

- (A)  $\frac{7n}{2}$  (B)  $\frac{7(n+1)}{2}$   
 (C)  $7n(n+1)$  (D)  $\frac{7n(n+1)}{2}$

195. If  $|Z^2 - 1| = |Z^2 + 1|$  then Z lies on

- (A) the real axis  
 (B) the imaginary axis  
 (C) a circle (D) an ellipse

196. The value of  $\frac{4(\cos 75^\circ + i \sin 75^\circ)}{0.4(\cos 30^\circ + i \sin 30^\circ)}$  is

- (A)  $\frac{\sqrt{2}}{10}(1+i)$  (B)  $\frac{\sqrt{2}}{10}(1-i)$   
 (C)  $\frac{10}{\sqrt{2}}(1-i)$  (D)  $\frac{10}{\sqrt{2}}(1+i)$

197. The value of  $\sum_{k=1}^{10} \left( \sin \frac{2k\pi}{11} + i \cos \frac{2k\pi}{11} \right)$  is

- (A) 1 (B) -1  
 (C) -i (D) i

198. If the roots of the quadratic equation  $x^2 + px + q = 0$  are  $\tan 30^\circ$  and  $\tan 15^\circ$  respectively, then the value of  $2 + q - p$  is

- (A) 0 (B) 1  
 (C) 3 (D) 2

199. If  $(1 - p)$  is a root of quadratic equation  $x^2 + px + (1 - p) = 0$ , then the roots are

- (A) 0, 1 (B) -1, 1  
 (C) 0, -1 (D) -1, 2

200. Solution of the inequation  $4^{-x+0.5} - 7 \cdot 2^{-x} < 4, x \in \mathbb{R}$  is

- (A)  $(-2, \infty)$  (B)  $(2, \infty)$   
 (C)  $\left(2, \frac{7}{2}\right)$  (D) None of these





# BIOLOGY

Question : 50

151. Neo-Darwinism believes that new species develop through
- (A) mutations with natural selection
  - (B) continuous variations with natural selection
  - (C) hybridisation
  - (D) mutations
152. Genetic drift operates in ..... population
- (A) small                      (B) large
  - (C) island                      (D) Mendelian
153. Which of the following is quartan in periodicity?
- (A) *P. ovale*                      (B) *P. vivax*
  - (C) *P. falciparum*                      (D) *P. malariae*
154. B.C.G. is vaccine against?
- (A) Typhoid
  - (B) Tuberculosis
  - (C) German measles
  - (D) Chicken pox
155. In tissue culture variations appeared are
- (A) Somatic variation
  - (B) Clonal variation
  - (C) Somaclonal variation
  - (D) Tissue culture variation
156. A common bio control agent for the control of plant diseases is
- (A) *Bacillus*                      (B) *Trichoderma*
  - (C) *Baculovirus*                      (D) *Glomus*



157. The technique for breakage of DNA fragment and inserting it into another DNA molecule, is related to  
 (A) Gene cloning  
 (B) Gene typing  
 (C) Gene splicing  
 (D) DNA fingerprinting
158. Which type of restriction enzymes are used in recombinant DNA technology?  
 (A) Type - I (B) Type - II  
 (C) Type - III (D) All of these
159. Which of the following bacteria has found extensive use in genetic engineering work in plants/best genetic vector used in plants?  
 (A) *Bacillus thuringiensis*  
 (B) *Xanthomonas citri*  
 (C) *Agrobacterium tumefaciens*  
 (D) *E. coli*
160. Animals with built in thermostat are  
 (A) poikilothermic (B) oligothermic  
 (C) homeothermic (D) biothermic
161. The lowest category in taxonomic hierarchy is  
 (A) class (B) kingdom  
 (C) species (D) phylum
162. Most primitive number in which roots are NOT present is  
 (A) Rhynia (B) Psilotum  
 (C) Lycopodium (D) Selaginella
163. Angiosperms differ from gymnosperms in having  
 (A) seeds (B) large leaves  
 (C) tap roots (D) covered seeds
164. Green glands are the excretory organs of  
 (A) Insecta (B) Myriapoda  
 (C) Arachnida (D) Crustaceans
165. Which tissue give mechanical strength to plant organs?  
 (A) Accessory cells (B) Collenchyma  
 (C) Parenchyma (D) Stomata
166. In which flower epipetalous stamen is found?  
 (A) *Calotropis* (B) *Sesbania*  
 (C) *Datura* (D) *Acalypha*
167. Which of the following is superficial of calf muscle?  
 (A) Trapezius (B) Latissimus  
 (C) Gluteus (D) Gastrocnemius
168. During inspiration in cockroach the respiratory passage is  
 (A) Stigmata  
 (B) Air chamber  
 (C) Spiracle and trachea  
 (D) Longitudinal respiratory tube
169. The function of the collateral gland in cockroach is to  
 (A) store eggs  
 (B) store sperms  
 (C) keep vagina moist  
 (D) secrete the egg case
170. Golgi apparatus takes part in  
 (A) Carbohydrate synthesis  
 (B) Lipid synthesis  
 (C) Protein synthesis  
 (D) Oxydative photophosphorylation
171. The longest living cells amongst the following are  
 (A) T-cells (B) B-cells  
 (C) Memory cells (D) RB
172. Mitochondria increases in the cells of  
 (A) dry seed  
 (B) dormant seed  
 (C) germinating seed  
 (D) Ripening fruits
173. What holds the ribosomes together in a polyribosome?  
 (A) mRNA  
 (B) rRNA  
 (C) tRNA  
 (D) mRNA, rRNA & tRNA





174. Some inorganic ions are required for enzyme activity. These inorganic substances are  
 (A) enzyme  
 (B) co-factor  
 (C) prosthetic group  
 (D) activator
175. Diploid chromosome number being 8, what shall be the number of chromatids in each daughter after Meiosis-I ?  
 (A) 2 (B) 4  
 (C) 8 (D) 16
176. Potassium ion exchange hypothesis of opening and closing of stomata was proposed by  
 (A) Sayre (B) Stewart  
 (C) Levitt (D) Bose
177. If a cell 'X' has  $op = 6$  and  $TP = 5$  and is surrounded by the cell with  $op = 4$  and  $TP = 2$ , then what will be the direction of water movement ?  
 (A) From other cell to cell 'X'  
 (B) From cell 'X' to other cell  
 (C) Water absorption is not affected by temperature  
 (D) Water will move freely
178. Bidirectional translocation of minerals takes place through  
 (A) xylem (B) phloem  
 (C) parenchyma (D) cambium
179. The intermediate between Glycolysis and TCA cycle is  
 (A) Oxaloacetate  
 (B) Glucose-1-6 diphosphate  
 (C) Pyruvic acid  
 (D) Acetyl Co-A
180. Out of 38 ATP molecules produced per glucose, 22 ATP molecules are formed from  $NADH/FASH_2$  in  
 (A) Respiratory chain  
 (B) Kreb's cycle  
 (C) Oxidative decarboxylation  
 (D) EMP
181. The maximum growth rate occurs in  
 (A) exponential phase  
 (B) lag phase  
 (C) stationary phase  
 (D) senescent phase
182. Mobilisation of stored food in germinating seed is triggered by  
 (A) Auxin (B) Cytokinin  
 (C) Gibberellin (D) Ethylene
183. Digestive enzymes are released by pancreas and bile is released by liver in response to the hormone  
 (A) Zymogen  
 (B) Cholecystokinin  
 (C) Insulin  
 (D) Secretin
184. After  $O_2$  diffusion into pulmonary capillaries, it diffused into ..... and binds with .....  
 (A) RBC, haemoglobin  
 (B) RBC,  $CO_2$   
 (C) Interstitial fluid,  $CO_2$   
 (D) Interstitial fluid, RBC
185. If vagus/parasympathetic nerve to heart is cut, the heart beat will  
 (A) stop (B) remain normal  
 (C) increase (D) decrease
186. Which of the following hormones causes reabsorption of  $Na^+$  and excretion of  $K^+$ ,  $H^+$  and  $H_2O$  ?  
 (A) LH (B) FSH  
 (C) TSH (D) Aldosterone
187. Which of the following animals having longitudinal binary fission ?  
 (A) Hydra (B) Plasmodium  
 (C) Paramoecim (D) Euglena
188. Vegetative propagation in mint occurs by  
 (A) Offset (B) Runner  
 (C) Sucker (D) Rhizome





189. Formation of an organism from a single, male gamete without fusion with egg is an example of  
 (A) Apogamy  
 (B) Parthenogenesis  
 (C) Parthenocarpy  
 (D) Apospory
190. Decrease in levels of which of the following causes menstrual flow?  
 (A) Progesterone (B) Vasopressin  
 (C) FSH (D) Oxytocin
191. Spermatozoa are nourished during development by  
 (A) Leydig cell (B) Sertoli cell  
 (C) Germinal epithelium  
 (D) Mitochondria
192. Genital warts STD is a viral disease and is caused by  
 (A) *Trichomonas vaginalis*  
 (B) *Treponema pallidum*  
 (C) *Human papilloma virus*  
 (D) *Chlamydia trachomatis*
193. The technique called gamete intrafallopian transfer (GIFT) is recommended for those females  
 (A) who cannot retain foetus inside uterus  
 (B) who cannot produce ovum  
 (C) who cannot provide suitable environment for fertilization  
 (D) whose cervical canal is too narrow, to allow passage for sperms
194. A method of birth control is  
 (A) IUDs (B) HJF  
 (C) IVF-ET (D) GIFT
195. The linked characters would always inherit together till they are  
 (A) mutated  
 (B) delinked due to segregation  
 (C) separated due to crossing over  
 (D) masked by dominance
196. Down's syndrome is a typical case of  
 (A) Nullisomy (B) Monosomy  
 (C) Gene mutation (D) Trisomy
197. Leading strand during DNA replication is formed  
 (A) in short segment  
 (B) continuously (C) first  
 (D) ahead of replication
198. An abnormal gene is replaced by normal gene. It is called  
 (A) Gene therapy (B) Cloning  
 (C) Mutation (D) None of these
199. Geographic limit within which a population exists is called  
 (A) Biome (B) Habitat  
 (C) Niche (D) Ecosystem
200.  $\frac{\Delta N_n}{\Delta N_t} = B$  represents  
 (A) Natality (B) Growth rate  
 (C) Mortality (D) All of these

