

**JEE-Main-08-04-2024 (Memory Based)**  
**[MORNING SHIFT]**

**Physics**

Question: In a LCR circuit at resonance, what will happen to current if resistance is halved?

Options:

- (a) Current will be Halved
- (b) Current will be doubled
- (c) Current will be tripled
- (d) Current will be quadrupled

Answer: (b)

Question: Identify the expression for Bernoulli's Equation

Options:

- (a)  $P + \rho gh + \frac{1}{2} \rho v^2 = \text{const}$
- (b)  $P^2 + \rho gh + \frac{1}{2} \rho v^2 = \text{const}$
- (c)  $P + \rho gh + \rho v^2 = \text{const}$
- (d)  $P + \frac{1}{2} \rho gh + \rho v^2 = \text{const}$

Answer: (a)

Question: A player caught a ball of mass 150 gm, travelling at 20 m/s. If the catching process was completed in 0.1 sec, find the force exerted by the ball on the player's hands.

Options:

- (a) 30 N
- (b) 60 N
- (c) 70 N
- (d) 90 N

Answer: (a)

Question: In a diffraction setup, if the slit width is  $b = 0.03$  cm and wavelength is 600 nm then the angular position of 2<sup>nd</sup> order minimum is  $\theta$ . Find the value of  $\sin \theta$

Options:

- (a) 0.004
- (b) 0.001
- (c) 0.002
- (d) 0.003

Answer: (a)

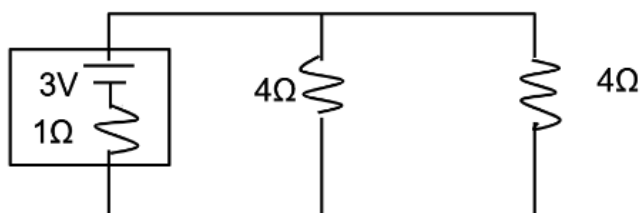
Question: Which of the following statements is INCORRECT regarding the paramagnetic substance ?

Options:

- (a) They have small but positive susceptibility
- (b) They are slightly attracted by magnetic field
- (c) Their susceptibility increases with rise in temperature
- (d) None of the above

Answer: (c)

Question: Find Potential difference across terminals of the cell



Options:

- (a) 1 V
- (b) 2 V
- (c) 3 V
- (d) 4 V

Answer: (b)

Question: Determine the ratio of specific heat at constant volume for monoatomic to Diatomic gas

Options:

- (a) 5/3
- (b) 3/5
- (c) 5/7
- (d) 7/5

Answer: (b)

Question: The resistance of conductor is 10 Ω at 0°C and 10.2 Ω at 100°C what Temp the resistance will be 10.95 Ω

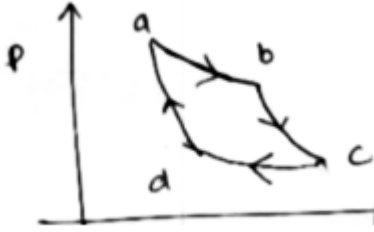
Options:

- (a) 234 K
- (b) 500 K
- (c) 748 K
- (d) 900 K

Answer: (c)

Question:

Options: The PV curve shown in the diagram consists of two isothermal and adiabatic curves. Then:



- (a)  $\frac{V_a}{V_d} = \frac{V_b}{V_c}$   
 (b)  $\frac{V_a}{V_d} = \left(\frac{V_b}{V_c}\right)^{-1}$   
 (c)  $\frac{V_a}{V_d} = \left(\frac{V_c}{V_b}\right)^2$   
 (d)  $\frac{V_a}{V_d} = \frac{V_c}{V_b}$

Answer: (a)

Question: In a series LCR Circuit, the value of resistance as well as  $(X_L - X_C)$  is halved, then the new current amplitude ( $I_2$ ) will satisfy: ( $I_1$  is old current amplitude)

Options:

- (a)  $I_2 = 2I_1$   
 (b)  $I_2 = 0$   
 (c)  $I_2 = \frac{I_1}{2}$   
 (d)  $I_2 = I_1$

Answer: (a)

Question: Proton and electron have same kinetic energy. Find ratio of their de-broglie wavelength.

Given  $m_p = 1836 m_e$

Options:

- (a)  $1 = \sqrt{1836}$   
 (b)  $\sqrt{1836} = 1$   
 (c)  $1 = 1836$   
 (d)  $1 : 1$

Answer: (a)

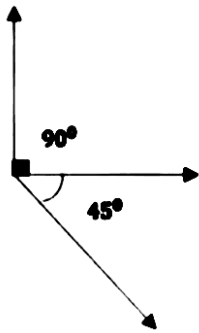
Question: Binding energy is  $18 \times 10^8$  Find Mass in Microgram

Options:

- (a) 10  
 (b) 20  
 (c) 30  
 (d) 40

Answer: (b)

Question: If the resultant of the vectors shown in  $A\sqrt{x}$ , find x.



Answer: (3)

Question: In a clock, second hand and minute hand are of 75 cm and 60 cm respectively. After 30 minutes, ratio of distance travelled by the tip of second hand to that of minute hand is x. Find x.

Options:

Answer: (75)

Question: Two planets are revolving around the sun of mass  $M_a$  and  $M_b$ . Ratio of angular momentum is 1:3. Find time period ratio in terms of masses.

Options:

(a)  $m_b^3/27m_a^3$

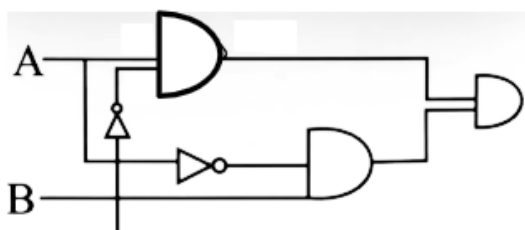
(b)  $m_a^3/27m_b^3$

(c)  $m_b^3/9m_a^3$

(d)  $m_a^3/9m_b^3$

Answer: (a)

Question: What would be the Output of the given logic gate?



Options:

(a)  $A \cdot B + \underline{A}$

(b)  $\underline{A} \cdot B + A$

(c)  $\underline{A} \cdot B + A + \underline{B}$

(d) 0

Answer: (d)

Question: The ratio of frequency of 7th overtone of closed organ pipe and open organ pipe of same length is \_\_\_.

Options:

(a) 15/6

- (b) 16/5
- (c) 1/1
- (d) 7/6

Answer: (a)

Question: Three masses  $M_A = 400$  gram,  $M_B = 1.2$  KG,  $M_C = 1.6$  KG have same. Kinetic Energy. Determine ratio of their linear momenta.

Options:

- (a) 1:  $\sqrt{5}$  : 2
- (b) 1:  $\sqrt{3}$  : 5
- (c) 1:  $\sqrt{3}$  : 2
- (d) 1:  $\sqrt{3}$  : 3

Answer: (c)

Question: If an area  $4\text{m}^2$  lies along  $2\sqrt{6}\hat{i} + 4\sqrt{6}\hat{j} + 2\sqrt{6}\hat{k}$

Find flux is  $\vec{E} = 4\hat{i} + 8\hat{j} + 4\hat{k}$

Options:

- (a) 3 Wb
- (b) 4 Wb
- (c) 5 Wb
- (d) 6 Wb

Answer: (d)

Question: Two conducting sphere of radii  $a$  and  $b$  are connected by wire. What is the ratio of their charges respectively?

Options:

- (a)  $a/b$
- (b)  $b/a$
- (c)  $ab$
- (d) None of these

Answer: (a)

Question: On a perfectly absorbing surface of area  $A$ , a light rays of intensity  $I = 360$  W/cm<sup>2</sup> falls normally if the force exerted is  $F$ , then find the area ( $A$  in m<sup>2</sup>)

Options:

- (a)  $F/12 \times 10^5$
- (b)  $F \times 10^5$
- (c)  $12F \times 10^5$
- (d) None

Answer: (a)

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**Chemistry**

Question: Which of the following do not undergoes disproportionation reaction?

Options:

- (a)  $F_2$
- (b)  $Cl_2$
- (c)  $Br_2$
- (d)  $I_2$

Answer: (a)

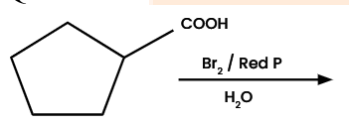
Question: Find the correct shape of the following molecules  $NH_3$ ,  $BrF_5$ ,  $PCl_5$ ,  $CH_4$

Options:

- (a)  $NH_3 \rightarrow$  Pyramidal ;  $CH_4 \rightarrow$  Square Pyramidal
- (b)  $BrF_5 \rightarrow$  Square Pyramidal ;  $CH_4 \rightarrow$  Tetrahedral
- (c)  $PCl_5 \rightarrow$  Trigonal Pyramidal ;  $BrF_5 \rightarrow$  Octahedral
- (d)  $NH_3 \rightarrow$  Pyramidal ;  $BrF_5 \rightarrow$  Tetrahedral

Answer: (b)

Question:

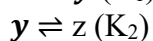
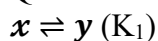


Options:

- (a)
- (b)
- (c)
- (d)

Answer: (c)

Question:





Find out equilibrium constant for  $x \rightleftharpoons w$ .

Options:

- (a)  $K_1 \times K_2 \times K_3$
- (b)  $K_1 + K_2 + K_3$
- (c)  $K_1 \times K_3$
- (d)  $K_1 \times K_2$

Answer: (a)

Question: If the wavelength of light is 3pm. Find out the frequency ?

Options:

- (a)  $10^{19} \text{ sec}^{-1}$
- (b)  $10^{20} \text{ sec}^{-1}$
- (c)  $10^{21} \text{ sec}^{-1}$
- (d)  $10^{18} \text{ sec}^{-1}$

Answer: (b)

Question: In the process of combustion of glucose ( $C_6H_{12}O_6$ ),  $CO_2$  and water formed, find amount of ( $O_2$ ) in g for complete combustion of (glucose).

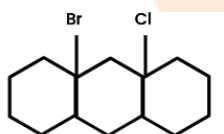
Molar mass of glucose (180 g/mol)

Options:

- (a) 32 g
- (b) 192 g
- (c) 16 g
- (d) 180 g

Answer: (b)

Question: Find out the Number of optical isomers?



Options:

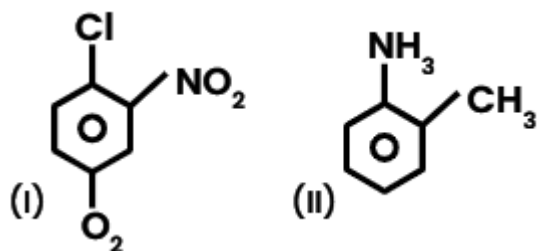
- (a) 15
- (b) 16
- (c) 32
- (d) 18

Answer: (b)

Question: Consider following statements.

Statement - 1 :- IUPAC name of (I) is 4-chloro-1,3-dinitrobenzene

Statement - 2 :- IUPAC name of (II) is 2-methylaniline



Options:

- (a) Both S-1 and S-2 are correct
- (b) S-1 is correct, S-2 is incorrect
- (c) S-1 is incorrect, S-2 is correct
- (d) Both S-1 and S-2 are incorrect

Answer: (c)

Question: We have two complexes,  $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$  and  $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$ , the magnetic properties respectively are

Options:

- (a) Diamagnetic and diamagnetic
- (b) Paramagnetic and Paramagnetic
- (c) Diamagnetic and Paramagnetic
- (d) Paramagnetic and Diamagnetic

Answer: (b)

Question: Find among the spin only magnetic moment (nearest integer) of M in  $\text{MO}^{2-}_4$ , M being the atom having least atomic radii among Sc, Ti, V, Cr, Mn, Zn.

Options:

- (a) 1
- (b) 2
- (c) 0
- (d) 3

Answer: (c)

Question: A solution contains 100 g water and 10 g of  $\text{AB}_2$ . The boiling of the solution was found to be 100.52%. The degree of dissociation of  $\text{AB}_2$  is :

[MW of Ab = 200 gm/mol ;  $K_b = 0.52 \text{ K kg/mol}$ ]

Options:

- (a) 0.5
- (b) 1
- (c) 2
- (d) 1.5

Answer: (a)

Question: Which of the following compounds will not give Hinsberg's test ?

Options:

- (a)  $\text{NH}_2 - \text{NH} - \text{CO} - \text{NH}_2$
- (b)  $\text{CH}_3\text{CO} - \text{NH}_2$
- (c)  $\text{CH}_3 - \text{CH}_2 - \text{NH}_2$
- (d)  $\text{CH}_3 - \text{NH} - \text{CH}_3$

Answer: (b)



Question: Statement - I : Stability of +1 oxidation state increases as  $Ga < In < Tl$ .

Statement - II : Stability of +1 oxidation state increases down the group due to inert pair effect.

Options:

- (a) Both S-1 and S-2 are correct
- (b) Both S-1 and S-2 are incorrect
- (c) S-1 is correct and S-2 is incorrect
- (d) S-1 is incorrect and S-2 is correct

Answer: (a)

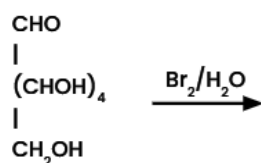
Question:  $CoCl_3 \cdot xNH_3$  on reaction with excess  $AgNO_3(aq.)$  gives  $tn$  mole of  $AgCl$  as precipitate. Summation of oxidation state of  $Co$  in  $CoCl_3 \cdot xNH_3$  and  $x$  is :

Options:

- (a) 8
- (b) 7
- (c) 9
- (d) 6

Answer: (d)

Question:



Options:

- (a)  $\begin{array}{c} \text{COOH} \\ | \\ (\text{CHOH})_4 \\ | \\ \text{COOH} \end{array}$
- (b)  $\begin{array}{c} \text{COOH} \\ | \\ (\text{CHOH})_4 \\ | \\ \text{CH}_2\text{OH} \end{array}$
- (c)  $\begin{array}{c} \text{CHO} \\ | \\ (\text{CHOH})_4 \\ | \\ \text{COOH} \end{array}$

(d) None of the above

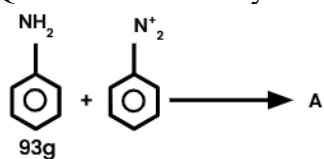
Answer: (a)

Question: Find out the magnitude of work done on the gas when 1 mole of an ideal gas undergoes compression from 9 litre to 1 litre through a reversible isothermal process. (in Joule) (Nearest integer)

Options:

Answer: (4980 J)

Question: How many moles of A will be formed ?



Options:

- (a) 2
- (b) 1
- (c) 4
- (d) 1.5

Answer: (b)

Question: Number of even number unpaired  $e^-$  is  $[\text{Co}(\text{NH}_2\text{O})_6]^{3+}$

Question: Match the following :-

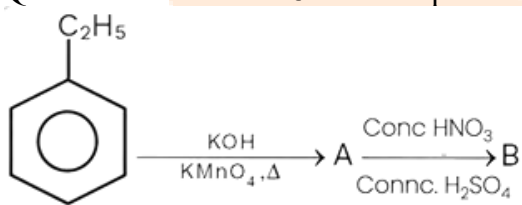
- |  |                    |
|--|--------------------|
| a. $[\text{Fe}(\text{SCN})]^{2+}$                                  | i. Yellow          |
| b. $[\text{Fe}(\text{CN})_5 \text{NOS}]^{4+}$                      | ii. Blood Red      |
| c. $\text{Fe}_4[\text{Fe}(\text{CN})_6] \cdot x\text{H}_2\text{O}$ | iii. Prussian Blue |
| d. $(\text{NH}_4)_3\text{PO}_4 \cdot 12\text{MoO}_3$               | iv. Purple         |

Options:

- (a) a  $\rightarrow$  ii, b  $\rightarrow$  iii, c  $\rightarrow$  iv, d  $\rightarrow$  i
- (b) a  $\rightarrow$  ii, b  $\rightarrow$  iv, c  $\rightarrow$  iii, d  $\rightarrow$  i
- (c) a  $\rightarrow$  i, b  $\rightarrow$  iii, c  $\rightarrow$  iv, d  $\rightarrow$  ii
- (d) a  $\rightarrow$  iii, b  $\rightarrow$  i, c  $\rightarrow$  ii, d  $\rightarrow$  iv

Answer: (b)

Question: Number of  $\pi$  bonds in product B



Options:

- (a) 5
- (b) 4
- (c) 3
- (d) 6

Answer: (a)

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**Maths**

**Question:** Find the range of  $\frac{\sin^4\theta + 3\cos^2\theta}{\sin^4\theta + \cos^2\theta}$   
**Answer:** [1, 3]

**Question:**  $A = \begin{bmatrix} 2 & a & 0 \\ 1 & 3 & 1 \\ 0 & 5 & b \end{bmatrix}$ ,  $A^2 = 4A^2 - A - 21I$  Find  $2a + 3b =$   
**Options:**

- (a) -10
- (b) 10
- (c) -13
- (d) 13

**Answer:** (c)

**Question:**  $\lim_{x \rightarrow 0} 2 \left[ \frac{1 - \cos x \sqrt{\cos 2x} \cdot \sqrt[3]{\cos 3x} \dots 10 \sqrt{\cos 10x}}{x^2} \right]$   
**Options:**

- (a) 55
- (b) 65
- (c) 56
- (d) 66

**Answer:** (a)

**Question:**  $\sin \theta = -\frac{3}{5}$  and  $\theta \in \left[ \pi, \frac{3\pi}{2} \right]$  then find  $80 (\tan^2 \theta - \cos \theta)$

**Answer:** (109)

**Question:** If  $I_n = \int_0^1 (1 - x^k)^n dx$  and if  $147I_{20} = 148I_{21}$ , find k.  
**Options:**

- (a) 4
- (b) 5
- (c) 6
- (d) 7

**Answer:** (d)

**Question:**  $\int \frac{6}{\sin^2 x (1 - \cot x)^2} dx$

**Question:** Find sum of solutions of the equation  $8^{2a} - 16 \cdot 8^a + 48 = 0$

**Question:**  $A = \begin{bmatrix} 2 & -1 \\ 1 & 1 \end{bmatrix}$ , Sum of diagonal elements of  $A^{13} = 3^n$  find n =  
**Options:**

- (a) 7

- (b) 8
- (c) 9
- (d) 10

**Answer: (7)**

**Question:**

In a hyperbola  $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ , eccentricity is  $\sqrt{3}$  Length of Latus rectum is  $4\sqrt{3}$ , If  $(\alpha, 6)$  Lies

on hyperbola. Product of focal distance from  $(\alpha, 6)$  is  $\beta$ , then  $\alpha^2 + \beta =$

**Options:**

- (a) 170
- (b) 171
- (c) 172
- (d) 173

**Answer: (b)**

**Question:**  $f(x) = 4\cos^3 x + 3\sqrt{3}\cos^2 x - 1$  then find the local maxima at point  $[0, 2\pi]$

**Question:** Let  $f(x) = \cos x - x + 1$ ,  $x \in [0, \pi]$ . Let  $M$  and  $m$  be the maximum and minimum values find  $(M-m)$

**Options:**

- (a)  $\pi$
- (b)  $\pi + 1$
- (c)  $\pi + 2$
- (d)  $2\pi$

**Answer: (c)**

**Question:** Let  $z$  be a complex number then  $|z + 2| = 1$  and

imaginary part of  $\frac{z+1}{z+2} = \frac{1}{6}$  then find the value of real part of  $z + 2$

**Question:**  $\vec{r}_1 = (5 + \mu)i + (1 - 3\mu)j + (1 + 2\mu)k$   
 $\vec{r}_2 = (2 + \lambda)i + (3 - 3\lambda)j + (3 + 4\lambda)k$  SD = ?

**Question:** Find 3 digit numbers using digits 0,2,4,6 and 7 without repetition and the number cannot be divisible by 3 is ?

**Options:**

- (a) 20
- (b) 24
- (c) 28
- (d) 30

**Answer: (c)**