## JEE Main 2024 Question Paper Feb 1 Shift 2 (B.E./B.Tech)

## JEE Main Physics Questions

Ques 1. Two trains run on North-South parallel tracks. Train A moves with velocity $20 \mathrm{~m} / \mathrm{s}$ towards North and train B moves with velocity $30 \mathrm{~m} / \mathrm{s}$ towards South. Then find the velocity of train B with respect to train A.

Ans. $50 \mathrm{~m} / \mathrm{s}$

Ques 2. A body of mass of 4 kg experiences two forces $\overrightarrow{\mathrm{F}_{1}}=5 \hat{\mathrm{i}}+8 \hat{\mathrm{j}}+$

$$
7 \hat{\mathrm{k}}, \& \overrightarrow{\mathrm{~F}}_{2}=3 \hat{\mathrm{i}}-4 \hat{\mathrm{j}}-3 \hat{\mathrm{k}} \text { then acceleration acting on the body } \mathrm{R}
$$

Ans. $\sqrt{ } 6$

Ques 3. A source produced electromagnetic wave of frequency 60 MHz . Find the wavelength of this wave in air.

Ans. 5 m

Ques 4. In the figure shown, find the ratio of tensions in the strings, $T_{1} / T_{2}$

3 kg
$\Delta T_{2}$

## 1 kg

A. $1 / 4$
B. $1 / 2$
C. $1 / 3$
D. 4

Ans. D

Ques 5. A Big drop is formed by coalescing 1000 small droplets of water. The surface water. The surface energy will become.

Ans. 1/10.

Ques 6. A solid sphere is rolling purely with speed von horizontal surface. It rolls up an incline surface and stops at height $h$. Then height $h$ is [ $g$ is the acceleration due to gravity]:
A. $3 \mathrm{v}^{2} / 10 \mathrm{~g}$
B. $7 \mathrm{v}^{2} / 10 \mathrm{~g}$
C. $5 v^{2} / 7 \mathrm{~g}$
D. $7 v^{2} / 5 \mathrm{~g}$

Ans. B

Ques 7. If the power of a light source is $P$ and frequency of photons emitted is $f$.
Find number photons emitted in time $t$.
A. $\mathrm{Pt} / 2 \mathrm{hf}$
B. $\mathrm{Pt} / \mathrm{hf}$
C. $1 \mathrm{pf} / 2 \mathrm{ht}$
D. Pf/ht

Ans. B

Ques 8. There are two cubical Gaussian surface carrying charges as shown. Find ratio of fluxes through surface $C_{1}$ and $C_{2}$ :

A. 1:1
B. 2:5
C. 5:2
D. 2:3

Ans. B

Ques 9. Find the number of significant digits in the value 10.05 :

Ans. 4

Ques 10. A ball of mass 120 g moving with initial velocity $25 \mathrm{~m} / \mathrm{s}$ is stopped by an external force $F$ in 0.15 sec . Find value of $F$ in newton :

Ans. 20

Ques 11. Find the ratio of the charge on $4 \mu \mathrm{~F}$ to that on $2 \mu \mathrm{~F}$ in steady state.


Ans. 3

Ques 12. If the rms velocity of hydrogen gas molecules is Vo, find the rms velocity of oxygen molecules at same temperature :
A. Vo
B. Vo/2
C. Vo/4
D. Vo/3

Ans. C

Ques 13. In the meter bridge shown below, the null point is at 40 cm from A. If $R$ is shunted by 22 , find the distance of new balance point from $A$

A. 22.7 cm
B. 60 cm
C. 62.5 cm
D. 60.5 cm

Ans. C

Ques 14. A particle is moving in circular path of radius $r$ speed $v$ such that speed is proportional to radius as $\mathrm{V} \propto \mathrm{r}^{3 / 2}$. Then how does time period of revolution depends on $r$ i.e $\mathrm{Tr}^{n}$ then n is.
A. $-1 / 2$
B. $5 / 2$
C. $-5 / 2$
D. $1 / 2$

Ans. B

Ques 15. In the given circuit, find electric current drawn from battery: In the given circuit, find electric current drawn from battery:

A. $3 / 4 \mathrm{~A}$
B. $4 / 3 \mathrm{~A}$
C. $4 / 5$
D. $5 / 4 \mathrm{~A}$

Ans. B

## JEE Main Chemistry Questions

Ques 1. Number of radial nodes present in 3p are
A. 0
B. 1
C. 2
D. 4

Ans. B

Ques 2. Which of the following compounds have colour due to d-d transition?
A. KMnO 04
B. K 2 Cr 2 O 7
C. K 2 CrO 4
D. CuSO4.5H2O

Ans. D

Ques 3. Which of the following compounds has intramolecular hydrogen bonding in it?
A. NH 3
B. H 20
C.

D.


Ans. B

Ques 4. Which of the following has the highest 3rd ionization energy?
A. Mn
B. V
C. Cr
D. Fe

Ans. A

Ques 5. A 10 mL hydrocarbon $\left(\mathrm{C}_{2} \mathrm{H}_{4}\right)$ on combustion give 40 mL CO 50 mL H20. Calculate the value of $\mathrm{x}+\mathrm{y}$

Ans. 14

Ques 6. Solubility of $\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ in 100 mL of pure water is W gm. Find out $\mathrm{K}_{\text {sp }}$ of $\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ is:
(M: Molecular mass of $\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ )
A. 108 * $(W / M)^{5}$
B. $108 * 10^{5} *[\mathrm{~W} / \mathrm{M}]^{5}$
C. $108 * 10^{4} *[\mathrm{~W} / \mathrm{M}]^{5}$
D. $108 * 10^{6} *[\mathrm{~W} / \mathrm{M}]^{5}$

Ans. B

Ques 7. Which of the following sets of elements can be detected by Lassaigne's Test?
A. N and S only
B. N, P and S only
C. P and halogens only
D. N, P, S and halogens

Ans. D

Ques 8. Which of the following compounds in 3d series does not show +3 oxidation state?
A. V
B. Cr
C. Mn
D. Cu

Ans. D

Ques 9. What is the order of reducing character for AsH3, NH3, PH3 (group 15 hydrides)?
A. NH3 > PH3 > AsH3
B. $\mathrm{PH} 3>\mathrm{NH} 3>\mathrm{AsH} 3$
C. AsH3> PH3 > NH3
D. NH3 > AsH3 > PH3

Ans. C

Ques 10. Which of the following compounds has the highest boiling point?
A. Butanol
B. Diethylether
C. Butane
D. Butanol

Ans. A

Ques 11. Consider the following two statements
Statement I: $\boldsymbol{\pi} 2 \mathrm{p}$ bonding molecular orbital has low electron density above \& below internuclear axis
Statement II: $\pi * 2 p$ antibonding molecular orbital has only one nodal plane
A. Both Statement I and Statement II are correct
B. Both Statement I and Statement II are incorrect
C. Statement I is incorrect, but Statement II is correct
D. Statement I is correct, but Statement II is incorrect

Ans. B

Ques 12. Consider the following two statements Statement I: $\mathrm{SiO}_{2}$ and $\mathrm{GeO}_{2}$ are acidic, SnO and PbO are amphoteric Statement II: Allotropes of carbon are formed due to catenation and dre $\rho \pi$ bond
A. Both Statement I and Statement II are correct
B. Both Statement I and Statement II are incorrect
C. Statement I is correct, but Statement II is incorrect
D. Statement I is incorrect, but Statement II is correct

Ans. C

Ques 13. Consider the following two statements
Statement I: In p and d block both metals and non - metals are present Statement II: Electronegativity and ionisation enthalpy of metals is greater than non-metals
A. Both Statement I and Statement II are correct
B. Both Statement I and Statement II are incorrect
C. Statement I is correct, but Statement II is incorrect
D. Statement I is incorrect, but Statement II is correct

Ans. B

Ques 14. Ethylene glycol of $x$ kg is mixed with 18.6 kg of solvent, $24^{\circ} \mathrm{C}$ depression in freezing point takes place.
Calculate value of x .
(Given: $\mathrm{K}_{1}=1.6^{\circ} \mathrm{C} / \mathrm{molal}$ M.W. of ethylene glycol $=\mathbf{6 2} \mathbf{~ g} / \mathrm{mol}$ )

Ans. 17

Ques 15. Find out charge (in C) required for electrolysis of 1 mole of $\mathrm{H}_{2} \mathrm{O}$ to produce $\mathrm{O}_{2}$ on one of the electrodes. ( $\mathrm{F}=96500 \mathrm{C}$ )

Ans. 193000

## JEE Main Mathematics Questions

Ques 1. Let $\alpha$ and $B$ the roots of equation $p x^{2}+q x-r=0$, where $P \neq 0$. If $p, q, r$ be the consecutive term of non constant G.P and $1 / \alpha+1 / \beta=$ $3 / 4$ then the value of $(\alpha-\beta)^{2}$ is:

Ans. 80/9

Ques 2. If the mirror image of the point $P(3,4,9)$ in the Line

$$
\frac{x-1}{3}=\frac{y+1}{2}=\frac{z-2}{1} \text { is }(\alpha, \beta, \gamma) \text { then find } 14(a+\beta+y) \text { is: }
$$

Ans. 108

Ques 3. The number of solution of the equation

$$
4 \sin ^{2} x-4 \cos ^{3} x+9-4 \cos x=0, x \in[-2 \pi, 2 \pi]
$$

is:

Ans. 0
Ques 4. If the domain of the function

$$
\begin{aligned}
& f(x)=\frac{\sqrt{x^{2}-25}}{\left(\sqrt{4-x^{2}}\right)}+\log \left(x^{2}+2 x-15\right) \text { is }(-\infty, \alpha) \cup(\beta, \infty) \text {, then } \\
& \alpha^{2}+\beta^{2} \text { is equal to } b
\end{aligned}
$$

Ans. 50

Ques 5. Let the system of equations $x+2 y+3 z=5,2 x+3 y+z=9$, $4 x+3 y+\lambda z=\mu$ have an infinite number of solutions. Then $\lambda+2 \mu$ is equal to

Ans. 17

Ques 6. The value of integrate $\int_{0}^{1}\left(2 x^{3}-3 x^{2}-x+1\right)^{1 / 3} d x$ is :
A. -1
B. 1
C. 0
D. 2

Ans. 0

Ques 7. The probability that Ajay will not go to office is $1 / 5$ and probability that Ajay and Vijay will not go to the office is $2 / 7$, if their visits to office is independent of each other, then find the probability that Ajay will go to the office, but Vijay will not go, is
A. 12/28
B. $13 / 35$
C. $18 / 35$
D. $24 / 35$

Ans. C

Ques 8. $\int_{0}^{\frac{\pi}{3}} \cos ^{4} x d x$ is equal to $a \pi+b \sqrt{ } 3$, then $a^{2}+b$ is equal to:
A. $1 / 2$
B. $1 / 8$
C. $1 / 4$
D. 1

Ans. B

Ques 9. Let $m$ and $n$ be the coefficient of 7 th and 13th term in expansion of $\left(1 / 3 x^{1 / 3}+1 / 2 x^{2 / 5}\right)^{18}$, then $(\mathrm{m} / \mathrm{n})^{1 / 3}$ is:
A. $1 / 4$
B. $4 / 7$
C. $1 / 9$
D. $4 / 9$

Ans. D

$$
\left|z+\frac{3+4 i}{2}\right|
$$

Ques 10. The minimum value of , $|z|<=1$ is,
A. $3 / 2$
B. $5 / 2$
C. 3
D. 5

Ans. A
Ques 11. Let vertex $A(2,3,1), B(3,2,-1), C(-2,1,3)$. If $A D$ is angle bisector of angle $A$, then projection of $\rightarrow A D$ on $\rightarrow A C$ is equal to:
A. $\sqrt{ } 3 / 2$
B. $\sqrt{ }(2 / 3)$
C. $\sqrt{ }(3 / 2)$
D. $2 / \sqrt{ } 3$

Ans. B
Ques 12. $d y / d x=\left(1-x-y^{2}\right) / y$ and $x(1)=1$, then $5 x(2)$ is equal to $\qquad$ .

Ans. 5
Ques 13. There are 20 lines numbered as $1,2,3, \ldots, 20$. And the odd numbered lines intersect at a point and all the even numbered lines are parallel. Find the maximum number of point of intersections

Ans. 101

Ques 14. Leta $a_{1}, a_{2}, a_{3}, \ldots, a_{n}$, be in A. P. and $S_{n}$ denotes the sum of first $n$ terms of this A. P. is $S_{10}=390, a_{10} / a_{50}=15 / 7$, then $S_{15}-S_{5}=$ $\qquad$ .

Ans. 365

