

MARKING SCHEME

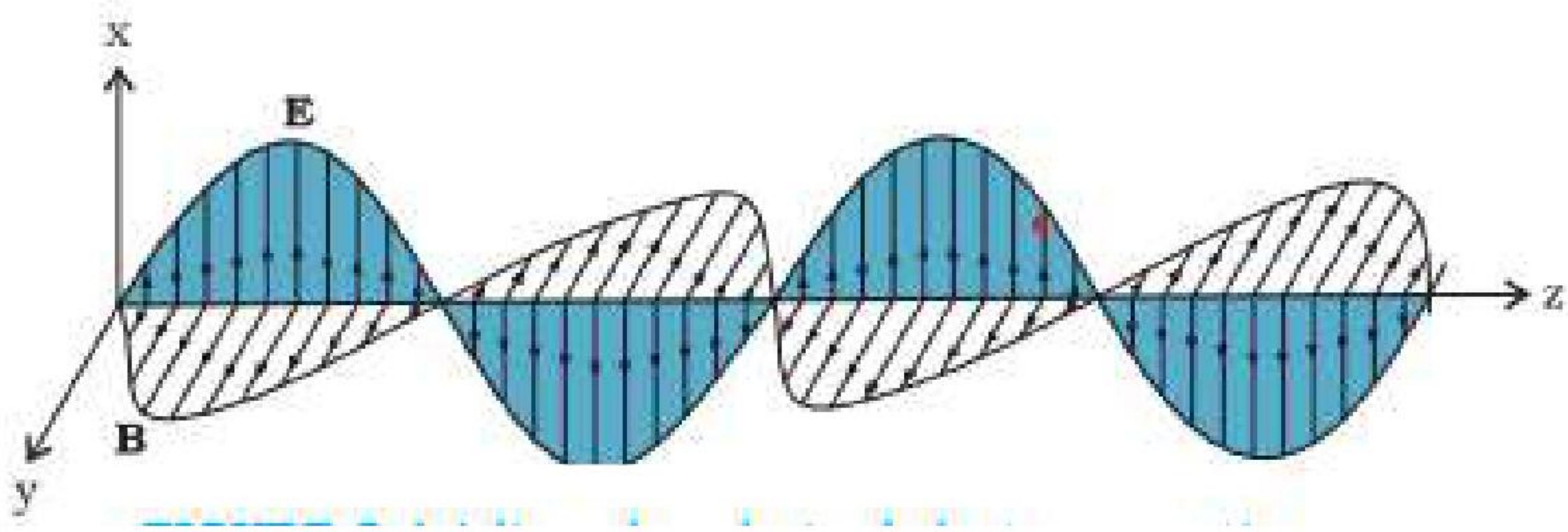
Senior Secondary School Term II Compartment Examination, 2022

PHYSICS (Subject Code-042)

[Paper Code: 55/6/3]

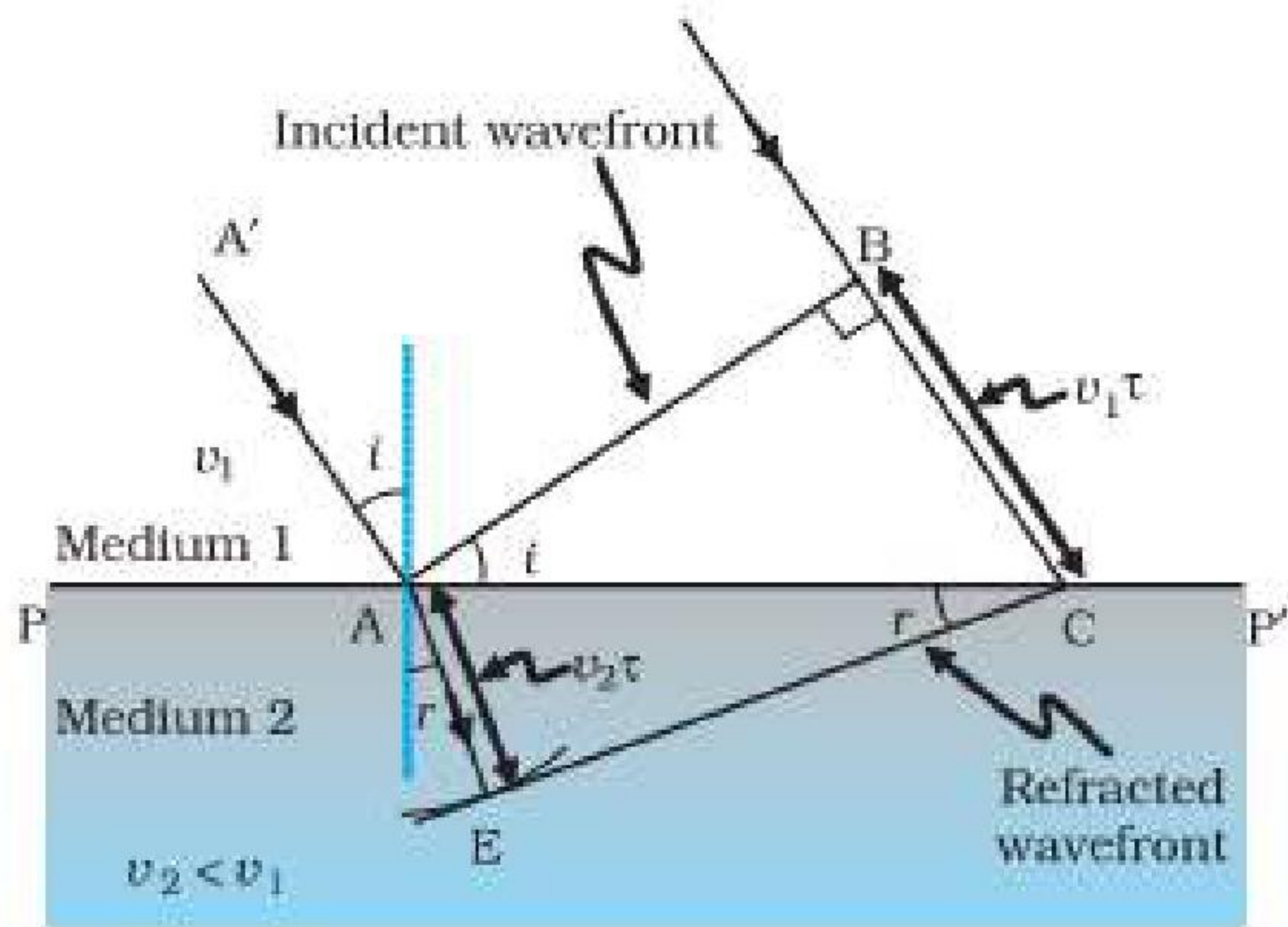
Q. No.	EXPECTED ANSWER / VALUE POINTS	Marks	Total Marks										
Section - A													
1	<table border="1" style="width: 100%;"> <tr> <td>a) Reason</td> <td style="text-align: right;">1</td> </tr> <tr> <td>b) Finding value of Kinetic energy</td> <td style="text-align: right;">$\frac{1}{2}$</td> </tr> <tr> <td>Finding value of potential energy</td> <td style="text-align: right;">$\frac{1}{2}$</td> </tr> </table> <p>a) Alpha particle reverses its direction of motion, due to strong repulsive force, exerted by positively charged nucleus without even actually touching the gold nucleus.</p> <p>b) $E_n = -3.4\text{eV}$ (given) $E_k = -E_n = 3.4\text{eV}$ $U = 2E_n = -6.8\text{eV}$</p>	a) Reason	1	b) Finding value of Kinetic energy	$\frac{1}{2}$	Finding value of potential energy	$\frac{1}{2}$	1 $\frac{1}{2}$ $\frac{1}{2}$	2				
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2	<table border="1" style="width: 100%;"> <tr> <td>(a) Justification</td> <td style="text-align: right;">1</td> </tr> <tr> <td>(b) Effect of closing one slit on fringe pattern in young's double slit Experiment</td> <td style="text-align: right;">1</td> </tr> </table> <p>(a) Energy required to liberate an electron (i.e ionisation energy) of C is greater than S_i. Hence number of free electrons for conduction in S_i is more than C.</p> <p>(b) When one of the slit is closed in young's double slit experiment than on the screen diffraction pattern is obtained. Or the fringe pattern will have a central bright maxima followed by secondary maxima of sharply decreasing intensity on both sides. Or the fringes are no longer equally spaced.</p>	(a) Justification	1	(b) Effect of closing one slit on fringe pattern in young's double slit Experiment	1	1 1	2						
(a) Justification	1												
(b) Effect of closing one slit on fringe pattern in young's double slit Experiment	1												
3.	<p>a)</p> <table border="1" style="width: 100%;"> <tr> <td>i) Reason of using shortwave band</td> <td style="text-align: right;">1</td> </tr> <tr> <td>ii) Reason of experimental demonstration in low frequency region</td> <td style="text-align: right;">1</td> </tr> </table> <p>i) Ionosphere reflects waves in these bands. ii) The frequency that we get even with modern electronic circuits is hardly about 10^{11} Hz, this is why experimental demonstration of electromagnetic waves had to come in low frequency region.</p> <p style="text-align: center;">OR</p> <p>b)</p> <table border="1" style="width: 100%;"> <tr> <td>i) Diagram</td> <td style="text-align: right;">1</td> </tr> <tr> <td>ii) Identification of wave</td> <td style="text-align: right;">$\frac{1}{2}$</td> </tr> <tr> <td>Use</td> <td style="text-align: right;">$\frac{1}{2}$</td> </tr> </table>	i) Reason of using shortwave band	1	ii) Reason of experimental demonstration in low frequency region	1	i) Diagram	1	ii) Identification of wave	$\frac{1}{2}$	Use	$\frac{1}{2}$	1 1	
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ii) Reason of experimental demonstration in low frequency region	1												
i) Diagram	1												
ii) Identification of wave	$\frac{1}{2}$												
Use	$\frac{1}{2}$												



	i) 	1	
	ii) X rays Use : As a diagnostic tool in medicine / treatment of cancer / or any other.	$\frac{1}{2}$ $\frac{1}{2}$	2

Section - B

4.	a) <table border="1" style="margin-left: auto; margin-right: auto;"> <tr> <td>a) Explanation</td> <td>1</td> </tr> <tr> <td>Two differences</td> <td>1 + 1</td> </tr> </table> <p>Maximum intensity is obtained at a point on the screen when the path difference is $(2n+1) \frac{\lambda}{2}$. With increasing 'n' only one-third, one-fifth, one-seventh etc of the slit contribute, hence intensity of maxima decreases sharply.</p> <p>Alternatively With the increase of order (n) the number of secondary wavelets responsible for the formation of secondary maxima, decreases resulting in sharp decrease of intensity.</p> <p>Differences (any two)</p> <ol style="list-style-type: none"> 1. Diffraction is a pattern formed as a result of superposition of waves from different portions of the same wave front. Interference is a pattern on a screen a result of superposition of single slit diffraction from two slits. 2. In diffraction pattern width of central maximum is twice the width of secondary maxima. In interference pattern width of each maxima is same. 3. In diffraction pattern intensity of maxima goes on decreasing as we move away from centre maximum. In interference pattern intensity of all maxima is same. 4. In diffraction pattern there is no absolute minima. In interference pattern absolute minima depends on amplitude of waves superposing. <p align="center">OR</p> <table border="1" style="margin-left: auto; margin-right: auto;"> <tr> <td>b) Diagram</td> <td>1</td> </tr> <tr> <td>Verification of Snell's law</td> <td>2</td> </tr> </table>	a) Explanation	1	Two differences	1 + 1	b) Diagram	1	Verification of Snell's law	2	1	1+1
a) Explanation	1										
Two differences	1 + 1										
b) Diagram	1										
Verification of Snell's law	2										



Consider the triangles ABC and AEC, we readily obtain

$$\sin i = \frac{BC}{AC} = \frac{v_1 \tau}{AC}$$

And

$$\sin r = \frac{AE}{AC} = \frac{v_2 \tau}{AC}$$

where i and r are the angles of incidence and refraction, respectively.

Thus we obtain

$$\frac{\sin i}{\sin r} = \frac{v_1}{v_2} \quad \text{-----(1)}$$

If c represents the speed of light in vacuum, then,

$$n_1 = \frac{c}{v_1}$$

and

$$n_2 = \frac{c}{v_2}$$

are known as the refractive indices of medium 1 and medium 2, respectively. In terms of the refractive indices, eq. (1) can be written as

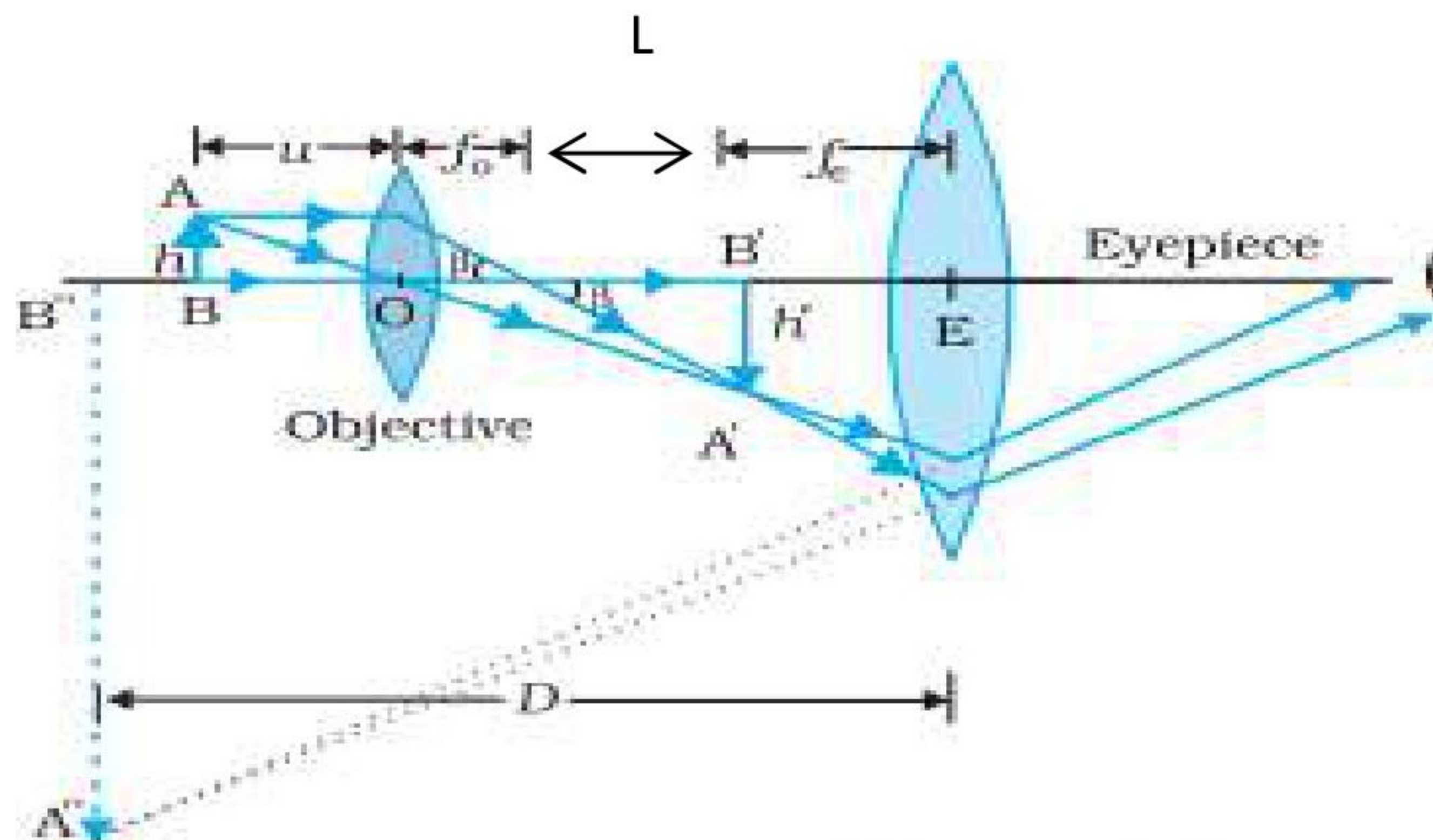
$$n_1 \sin i = n_2 \sin r$$

This is the Snell's law of refraction.

	<p>1</p> <p>1/2</p> <p>1/2</p> <p>1/2</p> <p>1/2</p> <p>3</p>	
5.	<div style="border: 1px solid black; padding: 5px;"> <p>a) Ratio of de- Broglie wavelengths and justification 1/2 + 1/2</p> <p>b) Identification of wavelengths and its justification 1/2 + 1/2</p> <p>Calculation of threshold frequency 1</p> </div> <p>a) de- Broglie wavelength, $\lambda = \frac{h}{mv}$</p> $\frac{\lambda_\alpha}{\lambda_p} = \frac{m_p}{m_\alpha}$ <p>As, $m_\alpha > m_p \Rightarrow \lambda_p > \lambda_\alpha$</p> <p>b) For photoelectric emission wavelength of radiation must be lesser than the threshold wavelength.</p> <p>Thus lights of wavelength 430 nm and 450 nm can cause photoelectric emission</p> <p>Threshold frequency (ν_0)</p> $\nu_0 = \frac{c}{\lambda_0} = \frac{3 \times 10^8}{600 \times 10^{-9}} = 5 \times 10^{14} \text{ Hz}$	<p>1/2</p> <p>1/2</p> <p>1/2</p> <p>1/2 + 1/2</p> <p>3</p>

6.

- | | | |
|----|--|---|
| a) | a) Ray diagram | 1 |
| | b) Derivation of angular magnification | 2 |



Linear magnification due to objective (m_o)

$$m_o = \frac{h'}{h} = \frac{L}{f_o} \text{----- (1)}$$

Linear magnification due to eye piece (m_e) when final image is formed at near point (D)

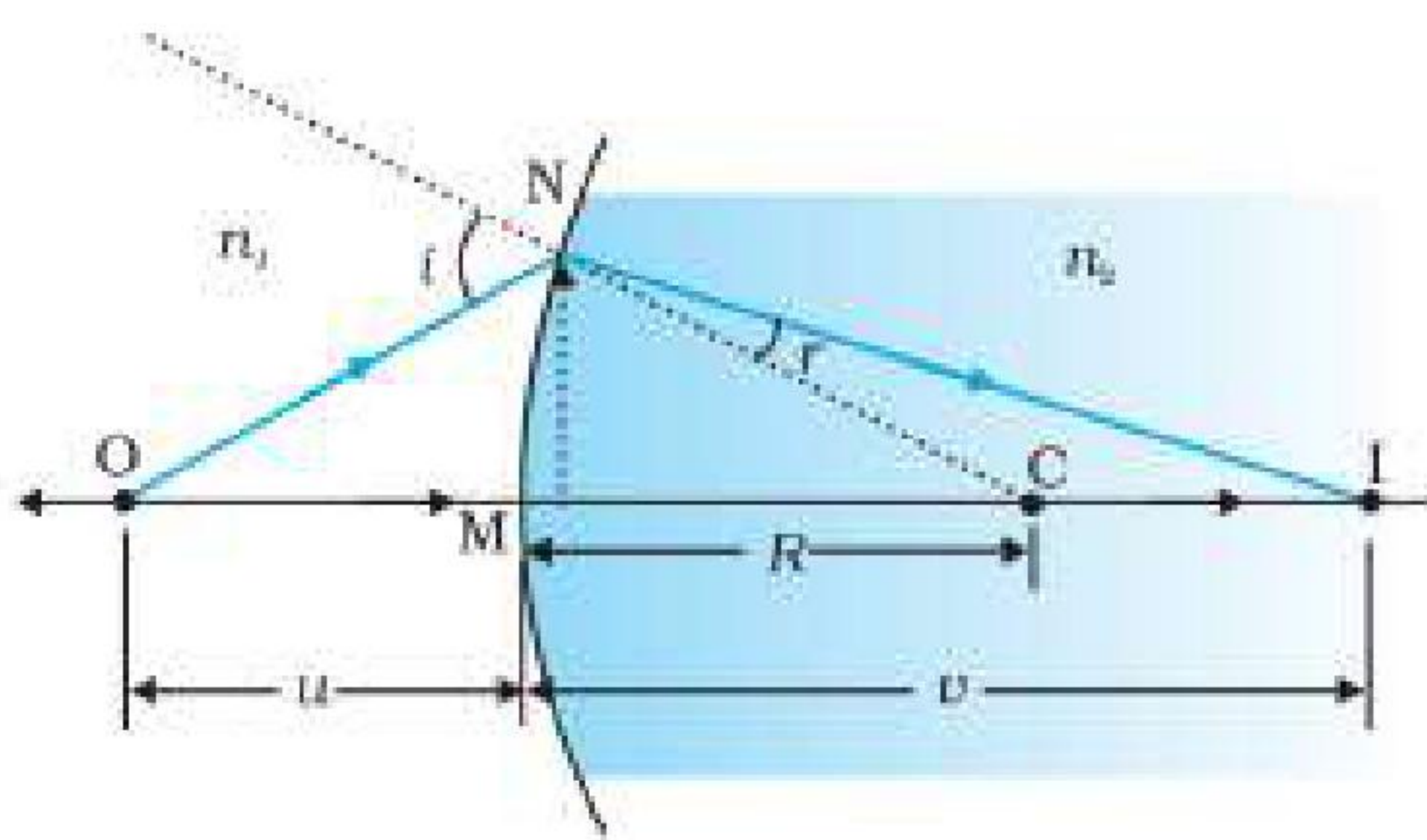
$$m_e = 1 + \frac{D}{f_e} \text{----- (2)}$$

Total magnification $m = m_o \times m_e$

$$m = \frac{L}{f_o} \left(1 + \frac{D}{f_e} \right)$$

OR

- | | | |
|----|----------------|---|
| b) | a) Ray diagram | 1 |
| | b) Derivation | 2 |



for small angles,
 $\tan \angle NOM = \frac{MN}{OM}$

$$\tan \angle NCM = \frac{MN}{MC}$$

$$\tan \angle NIM = \frac{MN}{MI}$$

Now, for ΔNOC , $\angle i$ is the exterior angle. Therefore, $\angle i = \angle NOM + \angle NCM$

$$i = \frac{MN}{OM} + \frac{MN}{MC}$$

Similarly,

1

1/2

1/2

1/2

1/2

1

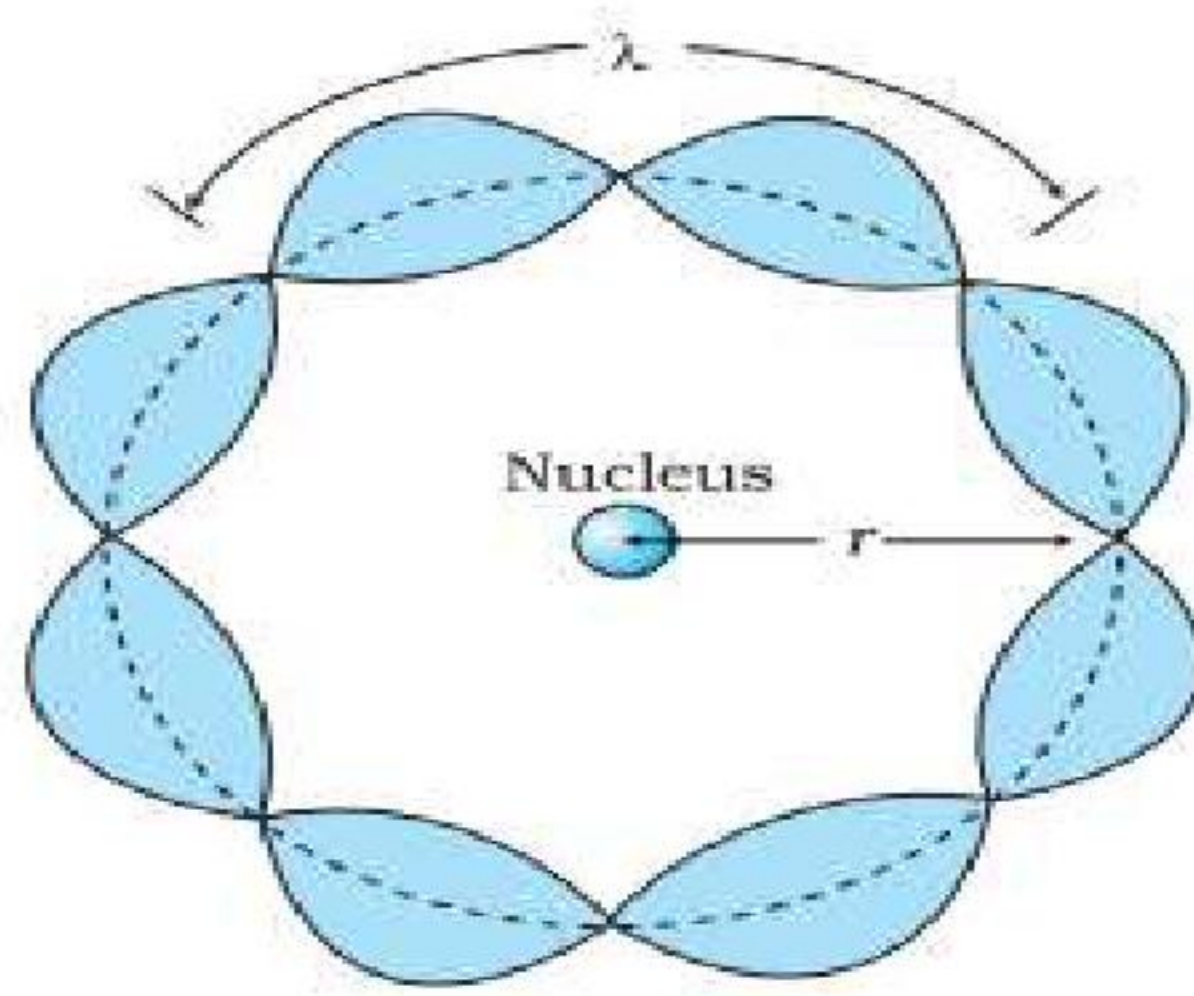
1/2



	<p>i.e. $r = \frac{MN}{MC} - \frac{MN}{MI}$ now, by Snell's law $n_1 \sin i = n_2 \sin r$ or for small angles, $\sin i \approx i$ and $\sin r \approx r$ $n_1 i = n_2 r$ Substituting i and r, we get $\frac{n_1}{OM} + \frac{n_2}{MI} = \frac{n_2 - n_1}{MC}$</p> <p>Here, OM, MI and MC represent magnitude of distances. Applying the Cartesian sign convention $OM = -u$, $MI = +v$, $MC = +R$ Substituting these in equation, we get $\frac{n_2}{v} - \frac{n_1}{u} = \frac{n_2 - n_1}{R}$</p>	<p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p>	3						
7.	<table border="1"> <tbody> <tr> <td>(a) Verifying the nature of the reaction</td> <td>2</td> </tr> <tr> <td>(b) Ratio of nuclear density</td> <td>1</td> </tr> </tbody> </table> <p>(a) mass of reactant = 226.02540 u mass of the products = $(222.01750 + 4.002603)u$ = $(226.020103)u$</p> <p>∴ Mass of reactants > Mass of products Hence the reaction is exothermic</p> <p>(b) 1:1</p>	(a) Verifying the nature of the reaction	2	(b) Ratio of nuclear density	1	<p>1</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p>	3		
(a) Verifying the nature of the reaction	2								
(b) Ratio of nuclear density	1								
8.	<table border="1"> <tbody> <tr> <td>(a) Finding ratio of Intensity of bright & dark fringes</td> <td>2</td> </tr> <tr> <td>(b) Effect of change in medium</td> <td></td> </tr> <tr> <td>(i) On energy carried</td> <td>$\frac{1}{2}$</td> </tr> </tbody> </table> <p>(a) $r = \frac{A_1}{A_2} = \sqrt{\frac{I_1}{I_2}} = \frac{2}{3}$</p> <p>$\frac{I_{\max}}{I_{\min}} = \left(\frac{r+1}{r-1}\right)^2 = \frac{25}{1}$</p> <p>(b) (i) No effect (c) (ii) Wavelength decreases</p>	(a) Finding ratio of Intensity of bright & dark fringes	2	(b) Effect of change in medium		(i) On energy carried	$\frac{1}{2}$	<p>1</p> <p>1</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p>	3
(a) Finding ratio of Intensity of bright & dark fringes	2								
(b) Effect of change in medium									
(i) On energy carried	$\frac{1}{2}$								
9.	<table border="1"> <tbody> <tr> <td>a) Obtaining Bohr's second postulate from de-Broglie hypothesis</td> <td>2</td> </tr> <tr> <td>b) Identification of transition of electron</td> <td>$\frac{1}{2} + \frac{1}{2}$</td> </tr> </tbody> </table>	a) Obtaining Bohr's second postulate from de-Broglie hypothesis	2	b) Identification of transition of electron	$\frac{1}{2} + \frac{1}{2}$				
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a)



For an electron moving in n^{th} circular orbit of radius r_n , the total distance is circumference of the orbit.

Thus $2\pi r_n = n\lambda$ (1) where $n = 1, 2, 3, \dots$

λ is de-broglie wavelength associated with the electron in the n^{th} orbit.

Now $\lambda = \frac{h}{p} = \frac{h}{mv_n}$ (2)

From equation (1) and (2)

$$2\pi r_n = \frac{nh}{mv_n}$$

$$mv_n r_n = \frac{nh}{2\pi}$$

Which is quantum condition proposed by Bohr for the angular momentum of the electron.

b) For Balmer series of hydrogen spectrum

$$\frac{1}{\lambda} = R \left(\frac{1}{2^2} - \frac{1}{n^2} \right)$$

(i) For maximum wavelength : transition of electron is from $n = 3$ to $n=2$

(ii) For minimum wavelength: transition of electron is from $n = \infty$ to $n=2$

$\frac{1}{2}$

$\frac{1}{2}$

$\frac{1}{2}$

$\frac{1}{2}$

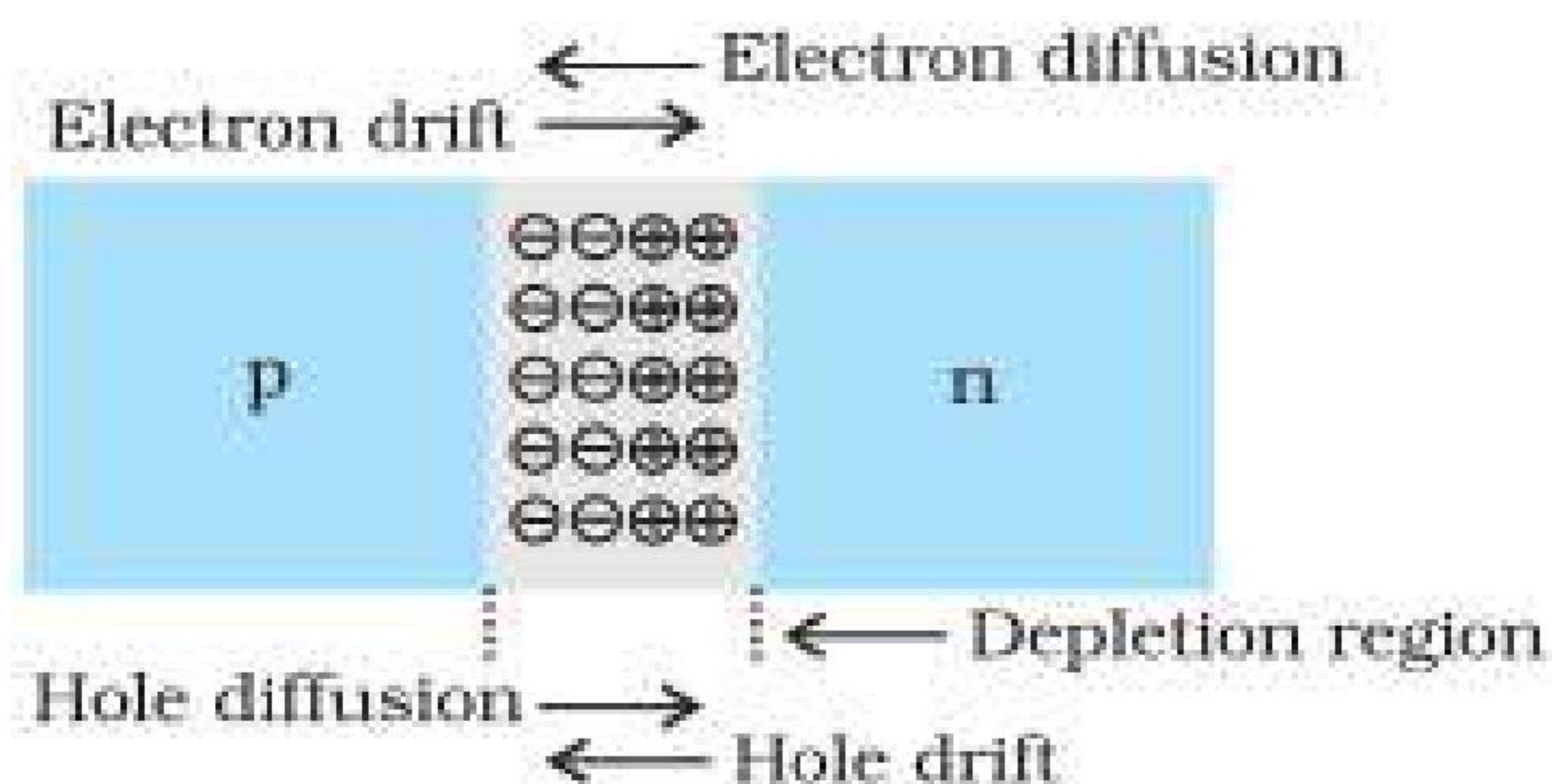
$\frac{1}{2}$

$\frac{1}{2}$

3

10.

- | | |
|---|----------------|
| a) Diagram | $\frac{1}{2}$ |
| Formation of p-n junction | $1\frac{1}{2}$ |
| b) Explanation of the need to join p and n type semiconductor at atomic level | 1 |



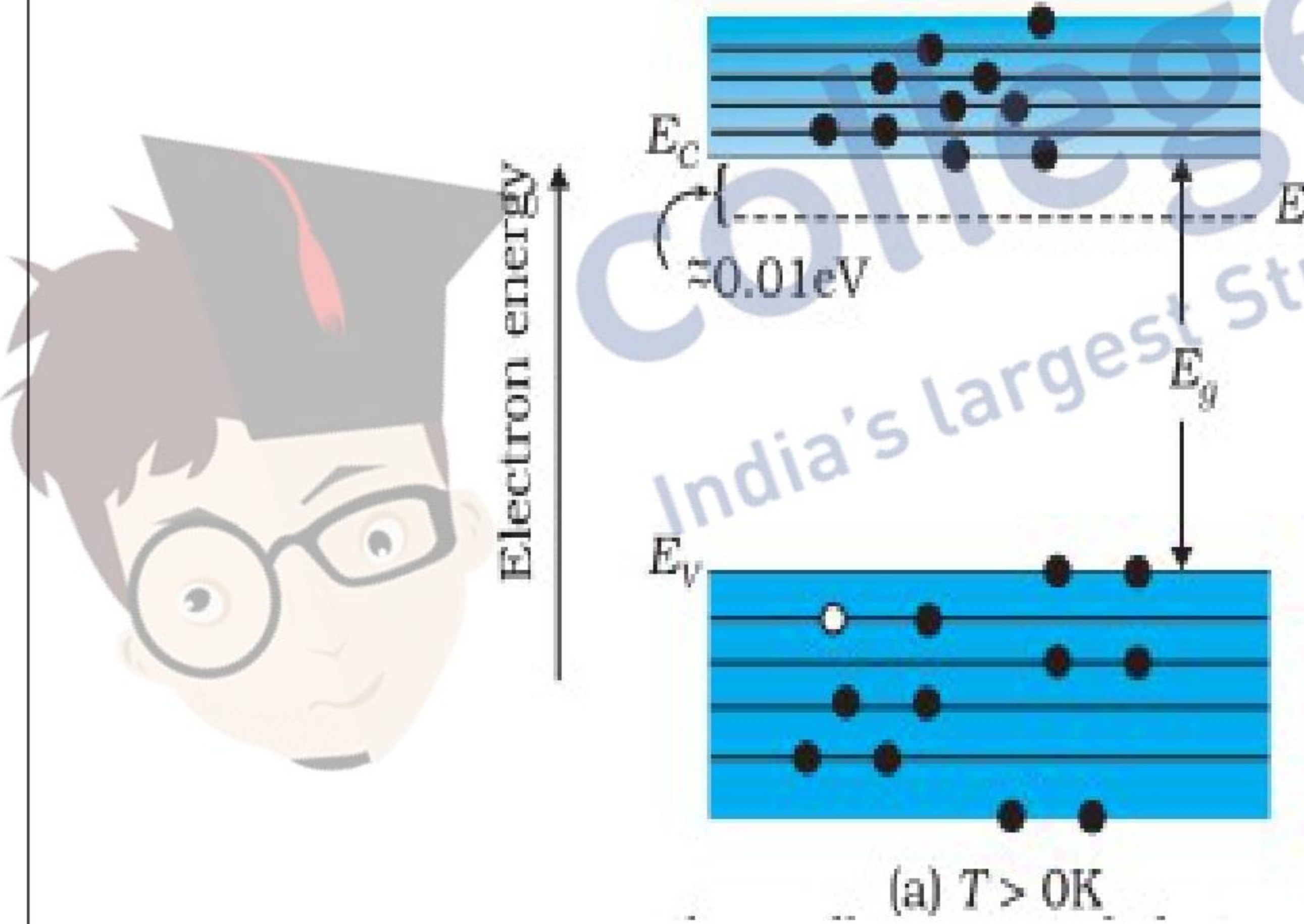
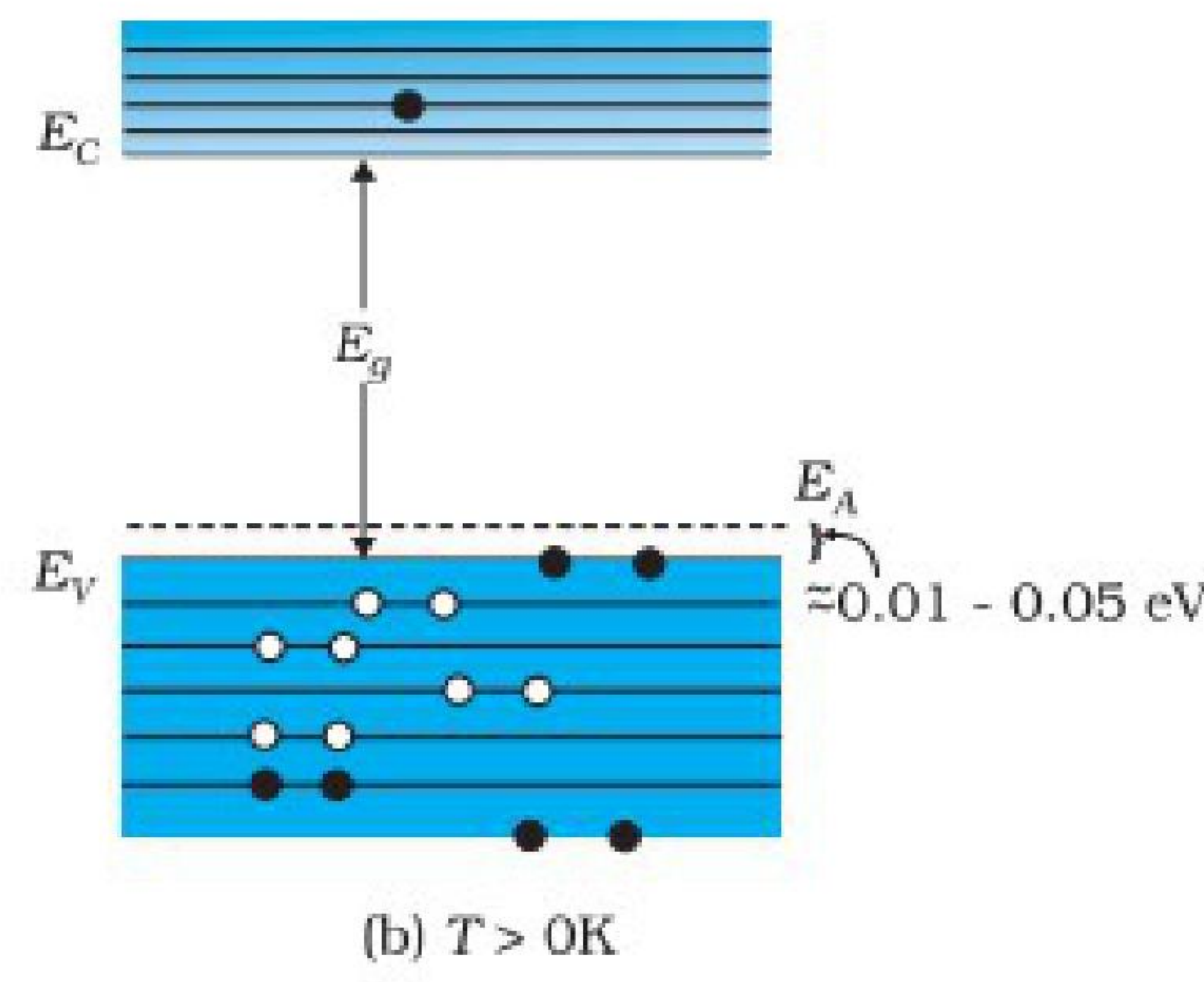
During the formation of p-n junction due to concentration gradient across p and n sides, holes diffuse from p side to n side and electrons diffuse from n side to p side. This motion of charge carriers gives rise to diffusion current across the junction.

$\frac{1}{2}$

$\frac{1}{2}$



	junction and diffusion of holes develops a layer of negative charge on p side of the junction. Due to this space charge region on either side of the junction an electric field is developed. This electric field drifts charge carriers across the junction and sets up drift current in a direction opposite to diffusion current. This process continues until the diffusion current is equal to drift current. Thus p-n junction is formed.	1/2	
	a) No, Any slab, howsoever flat, will have roughness much larger than the inter-atomic spacing (~2 to 3 Å) and hence continuous contact at the atomic level will not be possible. The junction will behave as a discontinuity for the flowing charge carriers.	1/2	
		1	3

11.	<table border="1"> <tr> <td>Diagram of energy band (at $T > 0K$)</td> <td></td> </tr> <tr> <td>n- type</td> <td>1</td> </tr> <tr> <td>p- type</td> <td>1</td> </tr> <tr> <td>Explanation</td> <td>1</td> </tr> </table> <p>Energy Band Diagram of n-type</p>  <p>(a) $T > 0K$</p> <p>Energy Band Diagram of p- type</p>  <p>(b) $T > 0K$</p> <p>In the energy band of n- type semiconductors, donor energy level E_D is</p>	Diagram of energy band (at $T > 0K$)		n- type	1	p- type	1	Explanation	1	1	1
Diagram of energy band (at $T > 0K$)											
n- type	1										
p- type	1										
Explanation	1										

	formed slightly below the bottom of E_C of the conduction band. Hence electrons from this level move into the conduction band easily. Note: If Students explain the cause by using the diagram itself give full credit to this part.	1	3
Section - C			
12.	a) (ii) b) (iv) c) (ii) d) (ii) e) (iv)	1 1 1 1 1	5



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