

**JEE-Mains-11-04-2023 [Memory Based]  
[Morning Shift]**

**Physics**

**Question:** Wave equation  $x = 10^{-2} \sin(160t - 0.2x + \pi/4)$  find wave velocity

**Options:**

- (a) 200 m/s
- (b) 400 m/s
- (c) 800 m/s
- (d) 1600 m/s

**Answer: (c)**

**Solution:**  $V = \frac{\omega}{K} = \frac{160}{0.2} = 800$

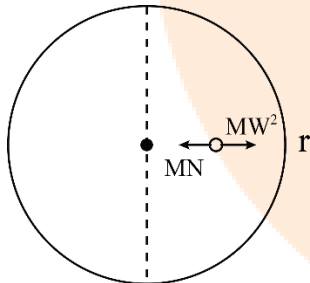
**Question:** Coin placed on a disc rotating with angular vel  $\omega$  slips at a distance 1 cm from centre. At what distance it will slip when angular velocity is halved?

**Options:**

- (a) 1 cm
- (b) 2 cm
- (c) 3 cm
- (d) 4 cm

**Answer: (d)**

**Solution:**



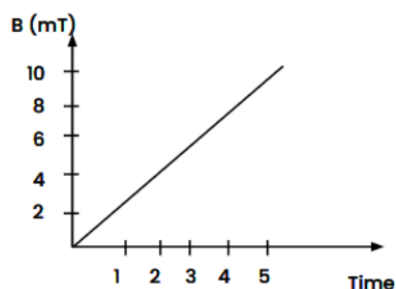
$$\mu N = m\omega^2 r$$

So

$$\frac{\mu N}{\mu N} = \frac{m\omega^2 (1)}{m\left(\frac{\omega}{2}\right)^2 .R}$$

$$x = 4 \text{ cm}$$

**Question:** If area of coil is  $4 \text{ m}^2$  find EMF from 2 to 4 seconds



**Options:**

- (a)  $8 \times 10^{-3}$  V
- (b)  $12 \times 10^{-3}$  V
- (c)  $4 \times 10^{-3}$  V
- (d)  $8 \times 10^{-2}$  V

**Answer: (a)**

**Solution:**

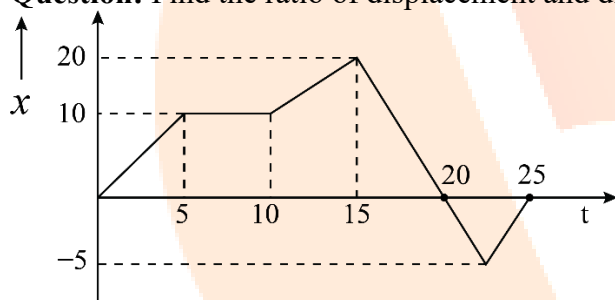
$$\varepsilon = \left| A \cdot \frac{dB}{dt} \right| \text{ and } \frac{dB}{dt} = 2 \left( \frac{mT}{s} \right)$$

So,

$$\varepsilon = 4 \times 2 \times 10^{-3} \text{ T/s}$$

$$\varepsilon = 8 \times 10^{-3} \text{ V}$$

**Question:** Find the ratio of displacement and distance.



**Solution:**

Positive area =

$$\left( \frac{1}{2} \times 5 \times 10 \right) + (5 \times 10) + \frac{1}{2} (30) 5 + \frac{1}{2} \times 5 \times 20$$

$$= 25 + 50 + 75 + 50 = 200$$

$$\text{Negative area} = \frac{1}{2} \times 5 \times 5 = \frac{25}{2}$$

$$\text{Distance} = \text{positive area} + \text{negative area} = 200 + \frac{25}{2} = \frac{425}{2}$$

$$\text{Displacement} = \text{positive area} - \text{negative area} = 200 - \frac{25}{2} = \frac{375}{2}$$

$$\text{So ratio} = \frac{15}{17}$$

**Question:** In photoelectric effect  $V_{\max,1} = V_0$  when wavelength is  $\lambda$  &  $V_{\max,2}$  is  $= V_0/4$ . When  $2\lambda$  is used. Find work function.

**Solution:**

$$\frac{1}{2} \times \left( eV_0 = \frac{hc}{\lambda} - \phi \right)$$

$$e \frac{V_0}{4} = \frac{hc}{2\lambda} - \phi$$

(-)

$$\frac{eV_0}{2} - \frac{eV_0}{4} = \frac{-\phi}{2} + \phi$$

$$\frac{eV_0}{4} = \frac{\phi}{2} \Rightarrow \phi = \frac{eV_0}{2}$$

**Question:** There are two identical resistances first they are joined in series then in parallel find ratio of heat produced in two cases assuming potential difference remains same

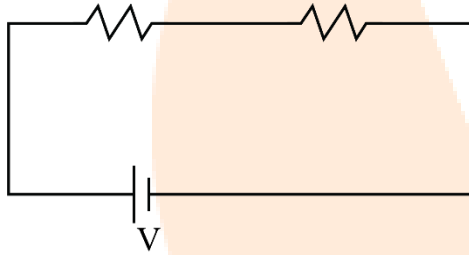
**Options:**

- (a) 3 : 1
- (b) 4 : 1
- (c) 5 : 1
- (d) 2 : 1

**Answer: (b)**

**Solution:**

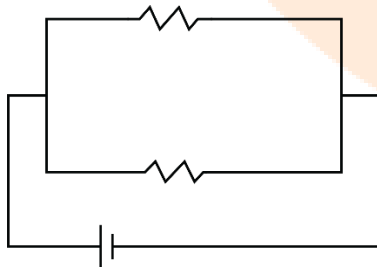
Case 1:



$$\text{Heat} = 2i^2Rt$$

$$= 2 \left( \frac{V}{2R} \right)^2 Rt = \left( \frac{V^2}{2R} \right) t$$

Case 2:



$$(\text{Heat})_2 = \frac{V^2}{(R/2)} t$$

$$= \frac{2V^2}{R} t$$

$$\text{So } \frac{(\text{Heat})_2}{(\text{Heat})_1} = \frac{2 \frac{V^2}{R} t}{\frac{V^2}{2R} t} = \frac{4}{1}$$

**Question:** A planet has density  $1/3$  of density of earth & radius 4 time that of earth. find ratio of acceleration due to gravity on surface of planet & on Earth

**Options:**

- (a) 4 : 3
- (b) 2 : 3
- (c) 5 : 7
- (d) 1 : 1

**Answer: (a)**

**Solution:**

$$g_s = \frac{GM}{R^2} = \frac{G\rho \frac{4}{3} \pi R^3}{R^2}$$

$$g_s = \frac{4}{3} \pi G \rho R$$

$$\text{So } \frac{g_p}{g_E} = \frac{\frac{4}{3} \pi G \left(\frac{1}{3}\right) 4}{\frac{4}{3} \pi G (1)(1)} = \frac{4}{3}$$

**Question:** A gun fires a bullet of mass 10 gm with vel 250 m/s as a result gun experiences a thrust force of 125 N find the number of bullets fired per second.

**Answer: 50.00**

**Solution:**

$$F = \left( \frac{NmU}{t} \right)$$

$$125 = \left( \frac{N}{t} \right) \cdot 10 \times 10^{-3} \times 250$$

$$\frac{N}{t} = \frac{1}{2} \times 100 = 50$$

**Question:** A force  $2 + 3x$  acts on particle at origin find work done from  $x = 0$  to  $x = 4$  (in J)

**Answer: 32.00**

**Solution:**

$$W = \int_{x_1}^{x_2} F dx = \int_0^4 (2 + 3x) dx = \left[ 2x + \frac{3}{2} x^2 \right]_0^4$$

$$= (2 \times 4) + \left( \frac{3}{2} \times 4^2 \right) - 0 = 8 + 24 = 32$$

**Question:** Current sensitivity is increased by 25% by changing N and A and keeping R constant. Find the change in voltage sensitivity

**Options:**

- (a) 25%
- (b) 50%
- (c) 12.5%
- (d) 75%

**Answer: (a)**

**Solution:**

$$\left(\frac{\phi}{V}\right) = \frac{1}{R} \times \left(\frac{\phi}{i}\right)$$

So equal increase if R is constant

So 25%.

**Question:** A projectile is fired at an angle  $30^\circ$  with horizontal, it has same speed at  $t = 3$  s and  $t = 5$  s. find initial velocity

**Options:**

- (a) 60 m/s
- (b) 70 m/s
- (c) 80 m/s
- (d) 90 m/s

**Answer: (c)**

**Solution:**

Velocity in X doesn't change and velocity magnitude in Y is also given same  $\theta = 30^\circ$

$$|V_x| = |V_y|$$

$$u \sin \theta - g(3) = |u \sin \theta - g(5)|$$

$$u \sin \theta - g3 = g5 - u \sin \theta$$

$$2u \sin \theta = 8g \Rightarrow u = 80$$

**Question:** Statement 1: au, parsec, lightyear are units of length.

Statement 2:  $\text{au} < \text{parsec} < \text{lightyear}$

**Options:**

- (a) S1 - Correct, S2 - Correct
- (b) S1 - Correct, S2 - False
- (c) S1 - False, S2 - Correct
- (d) S1 - False, S2 - False

**Answer: (b)**

**Question:** If height of transmitter antenna is 0 then find the receiver antenna in  $x \times 10^{-2}$  such that Line of sight is 4 km

**Answer: 125.00**

**Solution:**

$$\text{Range} = 0 + \sqrt{2R_E h_r}$$

$$4 \times 10^3 = \sqrt{2 \times 64 \times 10^5 \times x \times 10^{-2}}$$

$$16 \times 10^6 = 64 \times 2 \times 10^3 \times x$$

$$\frac{1000}{8} = x = 125$$

**Question:** Capacitor charged to potential V has energy  $U_1$  then it is connected to identical uncharged capacitor and final energy of system is  $U_2$  find  $U_2/U_1$

**Options:**

- (a) 1 : 1
- (b) 1 : 3

(c) 2 : 5

(d) 1 : 2

**Answer: (d)**

**Solution:**

$$U_1 = \frac{1}{2} \frac{Q^2}{C} \text{ and finally change divides equally}$$

$$U_2 = 2 \times \left[ \frac{1}{2} \frac{(Q/2)^2}{C} \right]$$

$$U_2 = \frac{1}{2} \left[ \frac{1}{2} \frac{QL}{C} \right]$$

$$\text{So } U_2 = \frac{1}{2} U_1 \Rightarrow \frac{U_2}{U_1} = 2$$

**Question:** Half life of nucleus A is equal to mean life of nucleus B find the relation between  $\lambda_A$  &  $\lambda_B$

**Options:**

(a)  $\lambda_A = \ln 2 / \lambda_B$

(b)  $\lambda_A = \ln 2 \lambda_B$

(c)  $\lambda_A = \lambda_B/2$

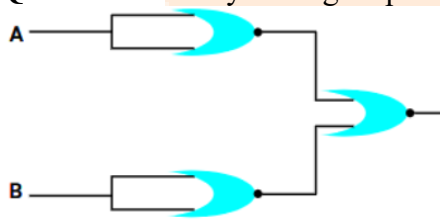
(d)  $\lambda_A = 2 \lambda_B$

**Answer: (b)**

**Solution:**  $\frac{\ln(2)}{\lambda_A} = \frac{1}{\lambda_B}$

$$\lambda_A = \ln 2 \lambda_B$$

**Question:** Identify the logic operation of following circuit.



**Options:**

(a) AND

(b) OR

(c) NOR

(d) NAND

**Answer: (a)**

**Question:** If the magnetic moment of both coils A & B are equal then choose the correct relation if  $r_A = 10$  cm,  $r_B = 20$  cm,  $N_A =$  Number of turns of coil A,  $N_B =$  Number of turns of coil B,  $I_A =$  Current in coil A,  $I_B =$  Current in coil B

**Options:**

(a)  $2N_A I_A = N_B I_B$

(b)  $N_A I_A = N_B I_B$

(c)  $N_A I_A = 4N_B I_B$

(d)  $N_A I_A = 2N_B I_B$

**Answer: (c)**

**Solution:**

$$M_1 = M_2$$

$$N_1 i_1 A_1 = N_2 i_2 A_2$$

$$N_1 i_1 \pi (10 \times 10^{-2})^2 = N_2 i_2 \pi (20 \times 10^{-2})^2$$

$$N_1 i_1 = 4 N_2 i_2$$

**Question:** If light is passing through a medium of critical angle  $45^\circ$  then wave speed is

**Options:**

(a)  $\frac{3}{\sqrt{2}} \times 10^8$

(b)  $3\sqrt{2} \times 10^8$

(c)  $\frac{3}{2} \times 10^8$

(d)  $3 \times 10^8$

**Answer: (a)**

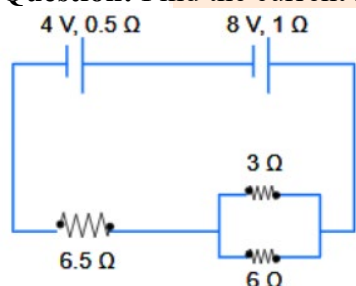
**Solution:**

$$\sin^{-1}\left(\frac{1}{\mu}\right) = i_c = 45^\circ$$

$$\frac{1}{\mu} = \frac{1}{\sqrt{2}} \Rightarrow \mu = \sqrt{2}$$

$$\text{So, } V = \frac{C}{\mu} = \frac{3 \times 10^8}{\sqrt{2}} \text{ m/s}$$

**Question:** Find the current flowing in  $3\Omega$  resistor in the given circuit.



**Options:**

(a) 0.4 A

(b) 0.2 A

(c) 0.8 A

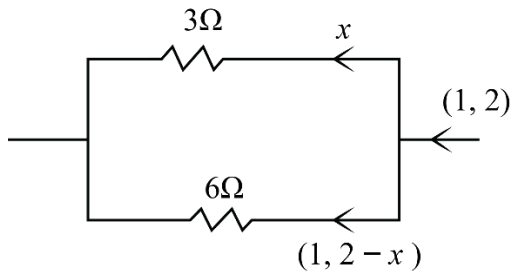
(d) 0.6 A

**Answer: (c)**

**Solution:**

$$i = \frac{12}{0.5 + 1 + 6.5 + 2} = 1.2$$

So,



$$3x = (1.2 - x)6$$

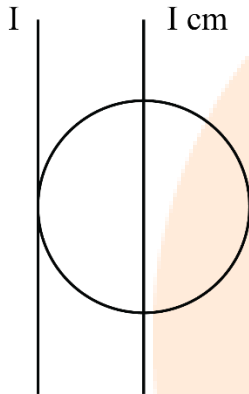
$$x = 2.4 - 2x$$

$$x = 0.8$$

**Question:** A solid sphere is rotating with  $\omega = 10 \text{ rad s}^{-1}$ . If  $I$  represents MOI about tangent to sphere &  $L$  represents angular momentum about diameter &  $I = [x \times 10^{-2}] L$ . Find  $x$  ?

**Answer: 35.00**

**Solution:**



$$I_{cm} = \frac{2}{5} mR^2$$

$$I = \frac{7}{5} mR^2$$

$$\text{So } \frac{I_{cm}}{I} = \frac{2}{7} \Rightarrow I_{cm} = \frac{2}{7} I$$

$$L = I_{cm} \omega = \frac{2}{7} I \omega$$

$$I = \frac{7}{2} \omega L = 35L$$

**Question:** For a scale, melting point is  $-15^\circ$ , boiling point is  $65^\circ$ , temp of  $-95^\circ$  on this scale will represent what value on fahrenheit scale?

**Options:**

- (a)  $100^\circ\text{F}$
- (b)  $-123^\circ\text{F}$
- (c)  $273^\circ\text{F}$
- (d)  $-148^\circ\text{F}$

**Answer: (d)**

**Solution:**



$$\frac{x - (-15)}{65 - (-15)} = \frac{F - 32}{180}$$

but  $x = -95$

$$\frac{-95 + 15}{80} = \frac{F - 32}{180}$$

$$-180 + 32 = F$$

$$\Rightarrow F = -148^\circ$$



## JEE-Mains-11-04-2023 [Memory Based] [Morning Shift]

### Chemistry

**Question:** Find the ratio of spin only magnetic moment of  $\text{Cr}(\text{CN})_6^{3-}$  and  $\text{Cr}(\text{H}_2\text{O})_6^{3+}$

**Options:**

- (a) 1 : 1
- (b) 1 : 2
- (c) 2 : 1
- (d) 2 : 3

Solution:  $\mu = \sqrt{n(n+2)}$

$\text{Cr} = 24(3d^5 4s^1), \text{Cr}^{+3} = 3d^3 4s^0$

Both  $\text{CN}^-$  and  $\text{H}_2\text{O}$  complex will have 3 unpaired electrons

$\therefore 1 : 1$

**Question:** Ionisation Energy order of period second elements?

Li, Be, C, B, N, O, F

**Options:**

- (a)  $\text{Li} > \text{Be} > \text{B} > \text{C} > \text{N} > \text{F} > \text{O}$
- (b)  $\text{F} > \text{O} > \text{N} > \text{C} > \text{Be} > \text{B} > \text{Li}$
- (c)  $\text{F} > \text{O} > \text{N} > \text{C} > \text{B} > \text{Be} > \text{Li}$
- (d)  $\text{F} > \text{N} > \text{O} > \text{C} > \text{Be} > \text{B} > \text{Li}$

**Answer:** (d)

**Solution:** Half filled and fully filled are extra stable

$\text{F} > \text{N} > \text{O} > \text{C} > \text{Be} > \text{B} > \text{Li}$

**Question: Statement-1:** Clean water would have BOD value of less than 5 ppm whereas highly polluted water could have a BOD value of 17 ppm or more.

**Statement-2:** For Clean water the maximum prescribed concentration of zinc and Nitrate is 5 ppm each.

**Options:**

- (a) Both statement 1 and statement 2 are correct
- (b) both statement 1 and statement 2 are incorrect
- (c) statement 1 is incorrect and statement 2 is correct
- (d) statement 1 is incorrect but statement 2 is correct

**Answer:** (a)

**Solution:** Clean water would have BOD value of less than 5 ppm whereas highly polluted water could have a BOD value of 17 ppm or more.

**Question:** Match the species with correct shape

Column-I	Column-II
(A) $\text{NH}_4^+$	(P) Bent
(B) $\text{ClO}_2^-$	(Q) Linear

(C) SF <sub>4</sub>	(R) Tetrahedral
(D) N <sub>3</sub> <sup>-</sup>	(S) See-saw

**Options:**

- (a) A – R; B – P; C – S; D – Q  
 (b) A – Q; B – P; C – R; D – S  
 (c) A – S; B – P; C – Q; D – R  
 (d) A – Q; B – P; C – S; D – R

**Answer: (a)**

**Solution:** Check hybridization for shapes

**Question:** Which of the following is correct set of non-ambidentate ligand?

**Options:**

- (a) NO<sub>2</sub>, EDTA  
 (b) SCN<sup>-</sup>, NO<sub>2</sub>  
 (c) C<sub>2</sub>O<sub>4</sub>, H<sub>2</sub>O  
 (d) SCN<sup>-</sup>, CN<sup>-</sup>

**Answer: (c)**

**Solution:** Fact based.

**Question:** 25% of 250g sugar solution & 40% of 500g sugar solution are mixed then find out the mass percentage in the solution

**Options:**

- (a) 35  
 (b) 45  
 (c) 40  
 (d) 38

**Answer: (a)**

**Solution:**  $25 \times 250 + 40 \times 500 = x \times 750$

$$x = 35$$

**Question:** 0.004M solution of K<sub>2</sub>SO<sub>4</sub> is isotonic with 0.01 M of glucose. Find degree of dissociation of K<sub>2</sub>SO<sub>4</sub>,

**Options:**

- (a) 75%  
 (b) 25%  
 (c) 50%  
 (d) 85%

**Answer: (a)**

**Solution:**  $\pi_1 = \pi_2$

For isotonic

$$\therefore 0.004 \times i \times RT = 0.01 RT$$

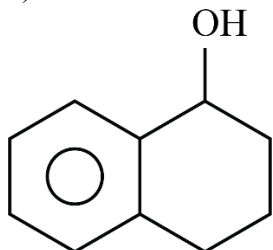
$$i = \frac{0.01}{0.004} = 2.5$$

$$\alpha = \frac{i-1}{n-1}$$

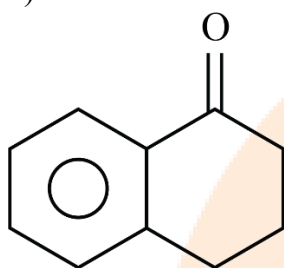
$$\alpha = 0.75$$

**Question:** Find correct order of electrophilic aromatic substitution reaction

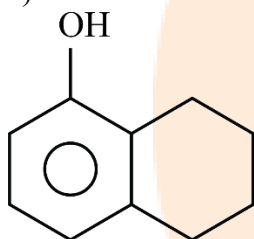
a)



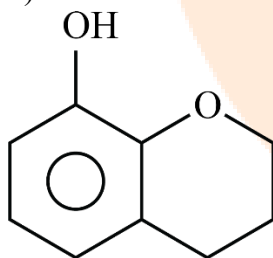
b)



c)



d)



**Options:**

(a) d > c > a > b

(b) c > d > b > a

(c) d > c > b > a

(d) a > b > c > d

**Answer: (a)**

**Solution:** EWG increases EAS reaction.

**Question: Statement-1:** CH<sub>4</sub> and H<sub>2</sub>O in presence of Ni Catalyst produces H<sub>2</sub> gas.

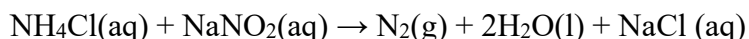
**Statement-2:** Sodium Nitrite reacts with NH<sub>4</sub>Cl gives H<sub>2</sub>, N<sub>2</sub> and H<sub>2</sub>O

**Options:**

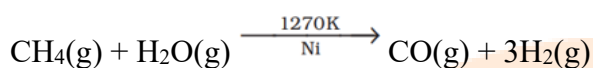
- (a) Both statement 1 and statement 2 are correct
- (b) Both statements are incorrect
- (c) Statement 1 is correct and statement 2 is incorrect
- (d) Statement 1 is incorrect and statement 2 is correct

**Answer: (a)**

**Solution:** In the laboratory, dinitrogen is prepared by treating an aqueous solution of ammonium chloride with sodium nitrate.



e.g.,



**Question:** If electrode potentials are

$$\text{Pb}^{2+}/\text{Pb} = m$$

$$\text{Pb}^{4+}/\text{Pb} = n$$

Then Find the value of x in?

$$\text{Pb}^{2+}/\text{Pb}^{4+} = m - xn$$

**Options:**

- (a) 2
- (b) 3
- (c) 4
- (d) 1

**Answer: (a)**

**Solution:**  $\Delta G_1^0 = -2Fm$ ,  $\Delta G_2^0 = -4Fn$

$$\Delta G_3^0 = \Delta G_1^0 - \Delta G_2^0 \text{ Thus } (m - 2n)$$

**Question:** To 25 ml of 1M silver nitrate 1.05 M potassium iodide is added dropwise. In colloidal Sol formed fixed and diffused layer consist of respectively:

**Options:**

- (a)  $\text{NO}_3^-$ ,  $\text{Ag}^+$
- (b)  $\text{Ag}^+$ ,  $\text{K}^+$
- (c)  $\text{Ag}^+$ ,  $\text{I}^-$
- (d)  $\text{Ag}^+$ ,  $\text{NO}_3^-$

**Answer: (d)**

**Solution:** Fixed layer of  $\text{Ag}^+$  on  $\text{AgI}$  and mobile layer of  $\text{NO}_3^-$

**Question:** A Solutions contain 2 salts anions,

**Statement-1:** Solution on treatment with freshly prepared  $\text{FeSO}_4$  gave a brown ring.

**Statement-2:** On reaction with  $\text{FeCl}_3$  and boiling gave a reddish-brown precipitate?

The two anions are :

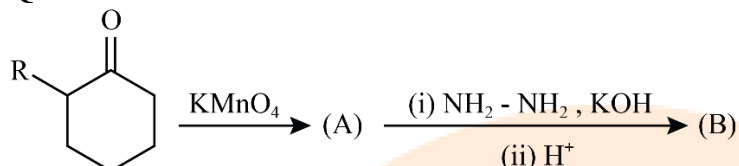
**Options:**

- (a) nitrate and acetate
- (b) nitrite and sulphate
- (c) nitrite and oxalate
- (d) nitrate and phosphate

**Answer: (a)**

**Solution:** Fact based

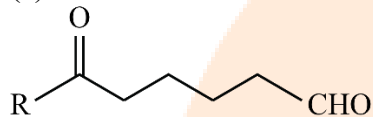
**Question:**



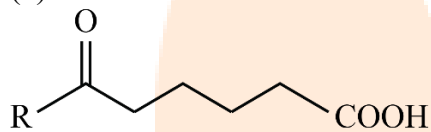
B is

**Options:**

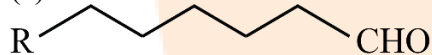
(a)



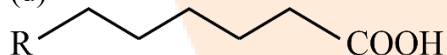
(b)



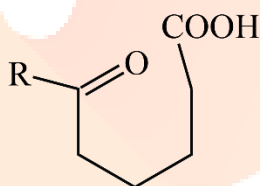
(c)



(d)



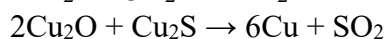
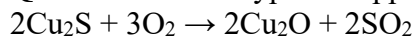
**Answer: (d)**



**Solution:** Popoff rule is followed to form A  
reduction of carbonyl is done

and Wolf Kishner's

**Question:** Which type of copper is formed by the following reactions?



**Options:**

- (a) Blister copper
- (b) Copper crisp
- (c) Reduced copper
- (d) Copper slag

**Answer: (a)**

**Solution:** Fact based

**Question:** Identify the correct statement about the compound  $\text{GaAlCl}_4$

**Options:**

- (a) Chlorine atom is bonded to both Ga and Al
- (b) Ga is cationic part and less electronegative than Al
- (c) Chlorine atom forms co-ordinate bond with Ga
- (d) Chlorine atom is bonded to Al

**Answer: (d)**

**Solution:** Fact based.

**Question:** In a container at constant temperature correct RMS velocity of the following.

**Options:**

- (a)  $\text{Ne} > \text{Cl}_2 > \text{UF}_6$
- (b)  $\text{Cl}_2 > \text{Ne} > \text{UF}_6$
- (c)  $\text{UF}_6 > \text{Ne} > \text{Cl}_2$
- (d)  $\text{UF}_6 > \text{Cl}_2 > \text{Ne}$

**Answer: (a)**

**Solution:**

$$u_{rms} = \sqrt{\frac{3RT}{M}}$$

$\therefore$  RMS is inversely proportional to molecular mass ( $\text{UF}_6 > \text{Cl}_2 > \text{Ne}$ )

**Question:** Which of the following can be represented as a meridional isomer?

**Options:**

- (a)  $[\text{Pt}(\text{NH}_3)_3\text{Cl}_3]^+$
- (b)  $[\text{Pt}(\text{en})_3]^{4+}$
- (c)  $[\text{Pt}(\text{en})_2\text{Cl}_2]^{2+}$
- (d)  $[\text{Pt}(\text{en})_2(\text{NH}_3)_2]^{4+}$

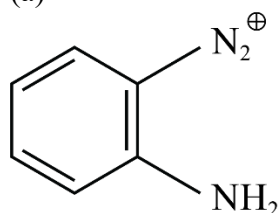
**Answer: (a)**

**Solution:** Fact based.

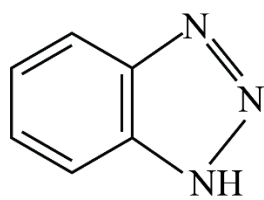
**Question:** o-phenylenediamine  $\xrightarrow{\text{HNO}_2}$

**Options:**

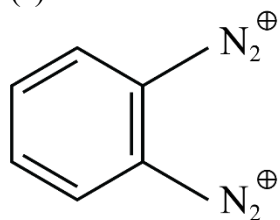
(a)



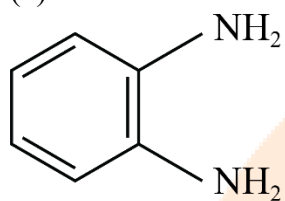
(b)



(c)

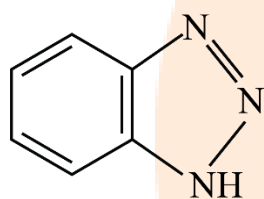


(d)



**Answer: (b)**

**Solution:**





**JEE-Mains-11-04-2023 [Memory Based]  
[Morning Shift]**

**Mathematics**

**Question:**  $z_1 = 5 + 2i$  is rotated by  $90^\circ$  anticlockwise to get  $\omega_1$ , and  $z_2 = 3 + i$  is rotated  $90^\circ$  clockwise to get  $\omega_2$ .  $\arg(\omega_1 - \omega_2) = ?$

**Answer:**  $\pi - \tan^{-1}\left(\frac{8}{3}\right)$

**Solution:**

$$\omega_1 = (5 + 2i)i = 5i - 2$$

$$\omega_2 = (3 + i)(-i) = -3i + 1$$

$$\omega_1 - \omega_2 = 8i - 3 = -3 + 8i$$

$$\begin{aligned} \arg(\omega_1 - \omega_2) &= \arg(-3 + 8i) \\ &= \pi - \tan^{-1}\left(\frac{8}{3}\right) \end{aligned}$$

**Question:** Number of integral terms in  $\left(3^{\frac{1}{2}} + 5^{\frac{1}{4}}\right)^{680}$  is

**Answer: 171.00**

**Solution:**

$$T_{k+1} = {}^{680}C_k (3)^{\frac{680-k}{2}} (5)^{\frac{k}{4}}$$

$$k = 0, 4, 8, \dots, 680$$

$$\text{Number of integral terms} = \frac{680}{4} + 1 = 171$$

So 171 terms

**Question:** In the expansion of  $(2 + x)^9$ , find the mean of the coefficients of  $x, x^2, x^3, \dots, x^7$ .

**Answer: 2736.00**

**Solution:**

$${}^9C_1 \times 2^8 + {}^9C_2 \times 2^7 + \dots + {}^9C_7 \times 2^2$$

$$\frac{3^9 - 2^9 - 9 \times 2 - 1}{7} = \frac{3^9 - 2^9 - 19}{7}$$

**Question:**  $M = (m_{ij}); m_{ij} \in \{0, 1, 2\}, i \geq 1, j \leq 2$ .  $A = \{m : M \text{ is invertible}\}$ . Number of  $n(A)$ .

**Answer: 52.00**

**Solution:**

Order:  $1 \times 1$ ,  $M = (1)$  or  $(2) \rightarrow 2$  matrices

Order:  $2 \times 2$ ,  $M = \begin{pmatrix} a & b \\ c & d \end{pmatrix} = ad - bc$

$ad = bc = 0$ , No. of matrices =  $5 \times 5 = 25$

$ad = bc = 1$ , No. of matrices =  $1 \times 1 = 1$

$ad = bc = 2$ , No. of matrices =  $2 \times 2 = 4$

$ad = bc = 4$ , No. of matrices =  $1 \times 1 = 1$

No. of invertible matrix of order  $2 \times 2$

$$= 3^4 - 25 - 1 - 4 - 1 = 50$$

Total =  $50 + 2 = 52$

**Question:** Five students given roll number and to be seated according to it. Find ways such that no students is on correct seat.

**Answer: 44.00**

**Solution:**

Number of seating arrangements

$$\begin{aligned} & 5! \left[ 1 - \frac{1}{1!} + \frac{1}{2!} - \frac{1}{3!} + \frac{1}{4!} - \frac{1}{5!} \right] \\ &= 5! \left( \frac{120 - 120 + 60 - 20 + 5 - 1}{120} \right) \\ &= 44 \end{aligned}$$

**Question:**  $a$  and  $b$  are the roots of  $x^2 - 7x - 1 = 0$ , then  $\frac{a^{21} + b^{21} + a^{17} + b^{17}}{a^{19} + b^{19}} = ?$

**Answer: 51.00**

**Solution:**

Given  $x^2 - 7x - 1 = 0$

$$a^2 - 1 = 7a$$

$$a^4 + 1 = 49a^2 + 2a^2$$

$$a^4 + 1 = 51a^2$$

$$\frac{a^{21} + b^{21} + a^{17} + b^{17}}{a^{19} + b^{19}} = \frac{a^{17}(51a^2) + b^{17}(51b^2)}{a^{19} + b^{19}} = 51$$

**Question:** Number of ordered truth values of  $(p, q, r)$  such that  $(p \vee q) \wedge (p \vee r) \rightarrow (q \vee r)$  is true.

**Answer: 7.00**

**Solution:**

False only if

$(p \vee q) \wedge (p \vee r)$  is T &  $q \vee r$  is F

$$q = r = F$$

So  $p$  has to be true

In remaining 7 cases it is True

**Question:**  $x + y + z = 15$ . Find number of ordered solutions  $(x, y, z)$  such that  $x, y, z \geq 0$

**Answer: 136.00**

**Solution:**

Total ordered solutions

$$\Rightarrow {}^{15+3-1}C_{3-1} = {}^{17}C_2$$

$$\Rightarrow {}^{15+2}C_2 = 136$$

**Question:** Area of region  $(x, y): x^2 + (y-2)^2 \leq 4; x^2 \geq 2y$  is

**Options:**

(a)  $\pi - \frac{8}{3}$

(b)  $\pi + \frac{8}{3}$

(c)  $2\pi - \frac{16}{3}$

(d)  $2\pi + \frac{16}{3}$

**Answer: (a)**

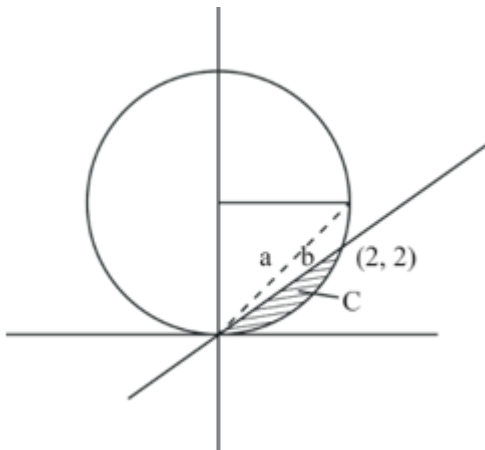
**Solution:**

$$x^2 = 2y$$

$$2y + (y - 2)^2 = 4$$

$$y^2 - 2y = 0$$

$$y = 0, 2$$



$$a + b + c = \frac{\pi}{4} \cdot 4 = \pi$$

$$a = \frac{1}{2} \cdot 2 \cdot 2 = 2$$

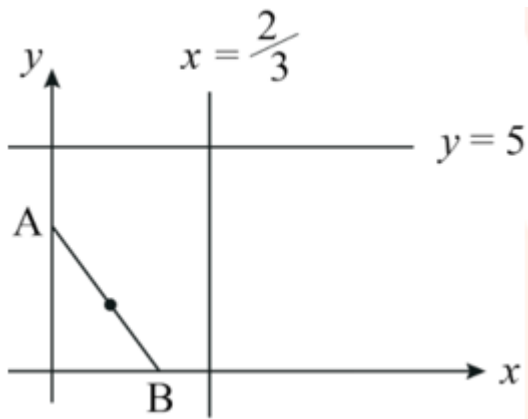
$$b = \frac{8}{6 \cdot 2} = \frac{2}{3}$$

$$c = \pi - 2 - \frac{2}{3} = \pi - \frac{8}{3}$$

**Question:** The lines  $y = 0$ ,  $x = 0$ ,  $y = 5$ ,  $x = \frac{2}{3}$  forms a rectangle.  $A(0, a)$  and  $B(b, 0)$  are points such that  $AB$  divides area of rectangle in  $1 : 4$ . Midpoint of  $AB$  lies on

**Answer:**  $xy = \frac{1}{3}$

**Solution:**



$$A_R = \frac{10}{3}$$

$$A_\Delta = \frac{ab}{2} = \frac{2}{3}$$

$$ab = \frac{4}{3}$$

$$\left(\frac{b}{2}, \frac{a}{2}\right) \text{ lies on } xy = \frac{1}{3}$$

**Question:** Number of elements in set  $S = [\theta \in [0, 2\pi] : 3\cos^4 \theta - 5\cos^2 \theta - 2\sin^6 \theta + 2 = 0]$  is

**Answer: 9.00**

**Solution:**

$$3\cos^4 \theta - 5\cos^2 \theta - 2\sin^6 \theta + 2 = 0$$

$$3\cos^4 \theta - 3\cos^2 \theta - 2\cos^2 \theta - 2\sin^6 \theta + 2 = 0$$

Writing in x terms

$$(3\cos^4 x - 3\cos^2 x) - 2\sin^6 x + 2\sin^2 x = 0$$

$$3\cos^2 x(\cos^2 x - 1) - 2\sin^2 x(-\sin^4 x + 1) = 0$$

$$3\cos^2 x \times \sin^2 x = 2\sin^2 x \times \cos^2 x(1 + \sin^2 x)$$

$$\sin^2 x = 0 \rightarrow 3 \text{ solutions}$$

$$\cos^2 x = 0 \rightarrow 2 \text{ solutions}$$

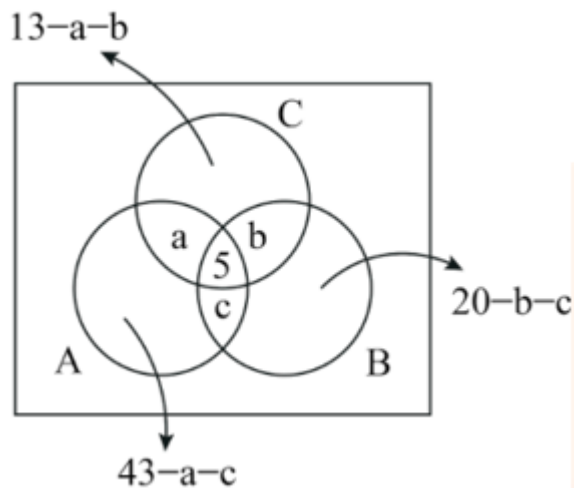
$$\sin^2 x = \frac{1}{2} \rightarrow 4 \text{ solutions}$$

So total 9 solutions.

**Question:** 48 medals were given in event A. 25 medals were given in event B and 18 medals were given in event C. Total men = 60. 5 men got medals in all 3 events. How many got medals in exactly 2 events.

**Answer:** 21.00

**Solution:**



$$76 - 2(a + b + c) + (a + b + c) + 5 = 60$$

$$\Rightarrow a + b + c = 21$$

**Question:**  $A^T = \alpha A - I$ ,  $|A^2 - A| = 4$ . Find sum of all possible values of  $\alpha$ . Order of A is 2.

**Answer:** 4.00

**Solution:**

For  $n = \alpha$ ,

$$A = \alpha A^T - I$$

$$A^T = \alpha A - I$$

$$A = \alpha(\alpha A - I) - I$$

$$A = \alpha^2 A - (\alpha + 1)I$$

$$A = \frac{-(\alpha + 1)}{1 - \alpha^2} I = \frac{I}{\alpha - 1}$$

$$|A^2 - A| = |A||A - I|$$

$$= \left(\frac{1}{\alpha - 1}\right)^n \left(\frac{2 - \alpha}{\alpha - 1}\right)^n = 4$$

$$(2 - \alpha)^2 = 4(\alpha - 1)^4$$

$$= 4[\alpha^4 - 4\alpha^3 + 1]$$

Sum of  $\alpha = 4$

**Question:**  $(1-x^2y^2)dx = xdy + ydx$ .  $y(1) = 2$ ,  $y(2) = \alpha$ . Find  $\alpha$ .

**Answer:**

**Solution:**

$$(1-x^2y^2)dx = d(xy)$$

$$\int dx = \int \frac{dxy}{1-x^2y^2}$$

$$x = \frac{1}{2} \ln \left| \frac{1+xy}{1-xy} \right| + C$$

$$(1, 2)$$

$$1 = \frac{1}{2} \ln |-3| + C$$

$$C = \frac{2}{\ln 3}$$

**Question:**  $f(x) = [x^2 - x] + [[x] - x]$  check continuity at  $x = 0, 1$

**Answer:** Discontinuous at  $x = 1$

**Solution:**

$$f(x) = [x^2 - x] + \{x\}$$

$$f(x) = [x^2 - x]$$

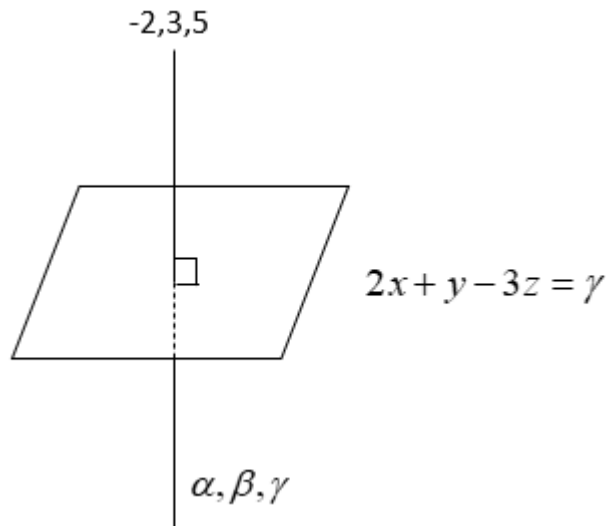
$$\lim_{x \rightarrow 0^+} f(x) = -1, \quad \lim_{x \rightarrow 0^-} f(x) = 0$$

$$\lim_{x \rightarrow 1^+} f(x) = 0, \quad \lim_{x \rightarrow 1^-} f(x) = -1$$

**Question:** Consider the plane  $2x + y - 3z = 6$ . If  $(\alpha, \beta, \gamma)$  is the image of point  $(-2, 3, 5)$  in the given plane, then  $\alpha + \beta + \gamma = \underline{\hspace{2cm}}$

**Answer:** 0.00

**Solution:**



$$\frac{\alpha + 2}{2} = \frac{\beta - 3}{1} = \frac{\gamma - 5}{-3} = \frac{-2(-22)}{14}$$

By solving

$$\alpha + \beta + \gamma = 0$$

**Question:**  $x_1, x_2, \dots, x_{100}$  are in A.P. If  $x_1 = 2$ ,  $\bar{x} = 200$ ,  $y_i = i(x_i - 1)$ ,  $i \in \{1, 2, \dots, 100\}$ , then  $\bar{y} = ?$

**Answer:**  $\frac{26765}{2}$

**Solution:**

$2, x_2, x_3, \dots, x_{100}$  are in AP

$$\frac{\frac{100}{2}(4 + 99 \times d)}{100} = 200$$

$$d = 4$$

$$y_i = i(2 + (i-1)d) - 1$$

$$y_i = i(2 + (i-1)4) - 1$$

$$y_i = i(4i - 3) = 4i^2 - 3i$$

$$\begin{aligned} \bar{y} &= \frac{4 \sum i^2 - 3 \sum i}{100} \\ &= \frac{4 \times \frac{100 \times 101 \times 201}{6} - 3 \times \frac{100 \times 101}{2}}{100} \\ &= \frac{26765}{2} \end{aligned}$$



**Question:** If  $\log_{\frac{x+7}{2}} \left( \frac{x+7}{2x+3} \right)^2 \geq 0$ , then total number of integral solutions is/are \_\_\_\_

**Answer: 7.00**

**Solution:**

**Case 1:** when base is  $0 < \left( x + \frac{7}{2} \right) < 1$  and  $\left| \frac{x+7}{2x+3} \right| \leq 1 \Rightarrow -1 \leq \frac{x+7}{2x+3} \leq 1$

$$(i) \frac{x+7}{2x+3} \leq 1 \Rightarrow \frac{x+7}{2x+3} \leq 1$$

$$x \in \left( -\infty, -\frac{3}{2} \right) \cup [4, \infty)$$

Or

$$(ii) -1 \leq \frac{x+7}{2x+3} \Rightarrow \frac{3x+10}{2x+3} \geq 0$$

$$x \in \left( -\infty, -\frac{10}{3} \right] \cup \left[ -\frac{3}{2}, \infty \right)$$

Required solutions for case 1  $x \in \left( -\frac{7}{2}, -\frac{5}{3} \right)$

**Case 2:** when base is  $\left( x + \frac{7}{2} \right) > 1$  and

$$\left| \frac{x+7}{2x+3} \right| \geq 1 \Rightarrow \frac{x+7}{2x+3} \geq 1 \text{ or } -1 \geq \frac{x+7}{2x+3}$$

$$(i) \frac{x+7}{2x+3} \geq 1 \Rightarrow \frac{x-4}{2x+3} \leq 0$$

$$x \in \left( -\frac{3}{2}, 4 \right)$$

Or

$$(ii) -1 \geq \frac{x+7}{2x+3} \Rightarrow \frac{3x+10}{2x+3} \leq 0$$

$$x \in \left( -\frac{10}{3}, -\frac{3}{2} \right) \cup \left( -\frac{5}{2}, -\frac{3}{2} \right)$$

Required solutions for case 2

$$x \in \left(-\frac{5}{2}, -\frac{3}{2}\right) \cup \left(-\frac{3}{2}, 4\right]$$

Total number of integral solutions =  $\pm 2, \pm 1, 0, 3, 4$

**Question:** Find the value of integral  $\int_{-\log e^2}^{\log e^2} e^x \left( \log_e \left( e^x + \sqrt{1+e^{2x}} \right) dx \right)$  is

**Answer:**

**Solution:**

$$\int_{-\log e^2}^{\log e^2} e^x \left( \log_e \left( e^x + \sqrt{1+e^{2x}} \right) dx \right)$$

Put  $e^x = t$

$$\int_{\frac{1}{2}}^2 \ln \left( t + \sqrt{1+t^2} \right) dt$$

$$t \ln \left( t + \sqrt{1+t^2} \right) \Big|_{\frac{1}{2}}^2 - \int_{\frac{1}{2}}^2 \frac{t}{\sqrt{1+t^2}}$$

$$2 \ln \left( 2 + \sqrt{5} \right) - \frac{1}{2} \ln \left( \frac{1+\sqrt{5}}{2} \right) - \frac{1}{2} 2 \times \sqrt{1+t^2} \Big|_{\frac{1}{2}}^2$$

$$2 \ln \left( 2 + \sqrt{5} \right) - \frac{1}{2} \ln \left( \frac{1+\sqrt{5}}{2} \right) - \left( \sqrt{5} - \frac{\sqrt{5}}{2} \right)$$